### POLLUTION SURVEY SUMMARY

### NORTH SASKATCHEWAN RIVER

1970 - 1971

by

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### SUMMARY

Water quality in the North Saskatchewan River deteriorated slightly this past winter as compared to the previous year, mainly due to lower discharge rates in the river.

Incidences of high Threshold Odor Number and Phenol concentration were noted at various locations but these did not result in any complaints from downstream users.

Values for bacteriological constituents remained high at locations immediately downstream from Edmonton. Coliform and E. Coli organisms exceeded guideline values in many instances at Fort Saskatchewan Bridge and Vinca Bridge.

New analyzing techniques were used to determine heavy metals. The results showed values of Mercury exceeding water quality objectives.

The lower flow rates experienced this past winter in the North Saskatchewan River are reflected on the concentration of other constituents such as Ammonia-Nitrogen and Phosphates. These nutrients continued to remain above guideline values despite decreased loadings to the river.

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### ESTABLISHMENT AND OPERATION OF THE PROGRAM

Many of the problems involved in the establishment and operation of the program are of practical rather than technical consideration. A summary discussion of some of the more basic considerations has been included.

#### PROGRAM OBJECTIVES

This program was established a number of years ago to provide basic information which would reveal pollution trends and provide water quality data for the many existing and prospective users of surface water in Alberta.

The monitoring program has these specific objectives:

- 1. To determine chemical, physical, bacteriological and biological characterisitics of Alberta's water under changing conditions.
- 2. To indicate, when possible, the sources of pollution entering a stream.
- 3. To compile data for future pollution abatement activities.
- 4. To determine background data on certain types of wastes, and to detect critical changes.
- 5. To obtain data useful for municipal, industrial, agricultural and recreational uses.
- 6. To procure data useful and necessary for securing public action toward the preservation of streams for all beneficial uses.

#### SCOPE OF ANALYSES

The frequency of sampling is admittedly limited. In some instances a more important parameter such as Dissolved Oxygen is monitored continuously. The routine analyses shown in this report represent the minimum of bacteriological, biochemical and chemical analyses which justify the program.

### SAMPLING STATIONS AND PROCEDURES

The following criteria were used as a basis for selection of the monitor stations.

- Waters used for municipal, industrial, agricultural, and recreational purposes.
- 2. Near known or suspected sources of pollution.
- 3. Near the Alberta-Saskatchewan border to determine the quality of water leaving the Province.

Stations were established at bridges whenever practical for convenient access to each watercourse. Routine samples included two litres for general chemical analyses, one litre each for Phenolics and Oil and Grease, and 100 millilitres in a sterile bottle for bacteriological analyses. Four litre samples were collected in a pyrex glass jar for Heavy Metals analyses. Four litre samples were also collected for Pesticides analyses.

#### NORTH SASKATCHEWAN RIVER

#### POLLUTION SURVEY

#### 1970 - 1971

#### INTRODUCTION

The North Saskatchewan River is used primarily for domestic and industrial uses, serving both as a water supply and as a receiving stream for waste water. In an attempt to control and maintain a reasonable quality of water in this river for downstream users, river sampling surveys are conducted throughout the year with special emphasis placed during the winter when pollutional effects are at a maximum. This report summarizes the efforts of the Division of Pollution Control in assessing the North Saskatchewan River water quality during the winter period of 1970-71.

Bi-weekly samples were taken at the following locations (see Fig. I):

- NS 1 Brazeau Reservoir Discharge
- NS 2 Drayton Valley Bridge
- NS 3 Devon Bridge
- NS 4 105 Street Bridge (Edmonton)
- NS 5 Fort Saskatchewan Bridge
- NS 6 Monitoring Station near Vinca Bridge
- NS 7 Waskatenau Bridge
- NS 8 Duvernay Bridge
- NS 10 Canadian Salt Company Water Intake
- NS 11 Lloydminster Ferry

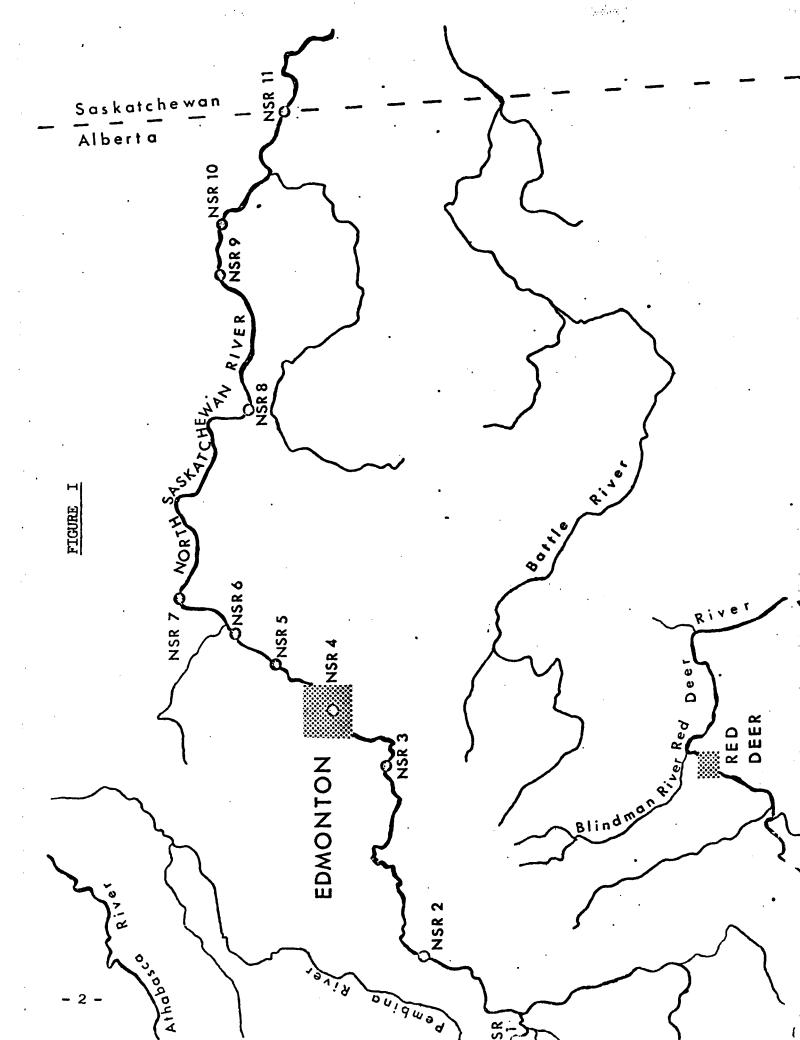
The four upstream locations from Edmonton, including Edmonton's 105th

Street Bridge, provide the background levels of constituents in the river.

Fort Saskatchewan Bridge is the first location where the quality of water
is influenced by major waste discharge sources. Vinca Bridge, Waskatenau

Bridge, Duvernay Bridge, Canadian Salt Company Water Intake, and Lloydminster

Ferry locations indicate the quality of the water and performance of the



river as it moves from a zone of decomposition of degradable material through a zone of recovery.

Aside from manual sampling of the river, an automatic water monitor was placed into operation at the Vinca Bridge site on December 15, 1970. The "Honeywell Water Quality Data Acquisition System" operated a total of 112 days of continuous operation until April 6, 1971. Dissolved Oxygen, pH, Temperature, Conductivity, and Turbidity were the parameters measured by the monitor. Significant variations in some of the parameters were observed during the winter indicating "slug release" of waste water upstream (see Fig. II). In all instances these recorded variations in water quality were traced to accidental spills.

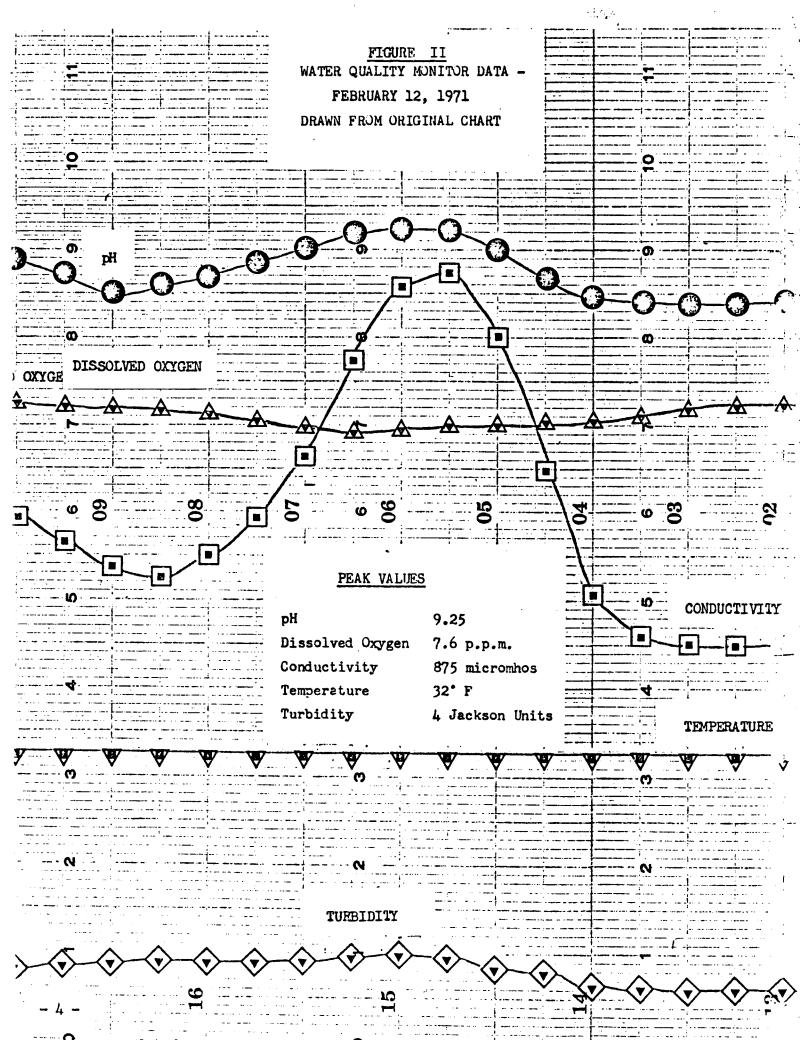
Industrial and municipal discharges to the North Saskatchewan River from all major sources were sampled in conjunction with river surveys.

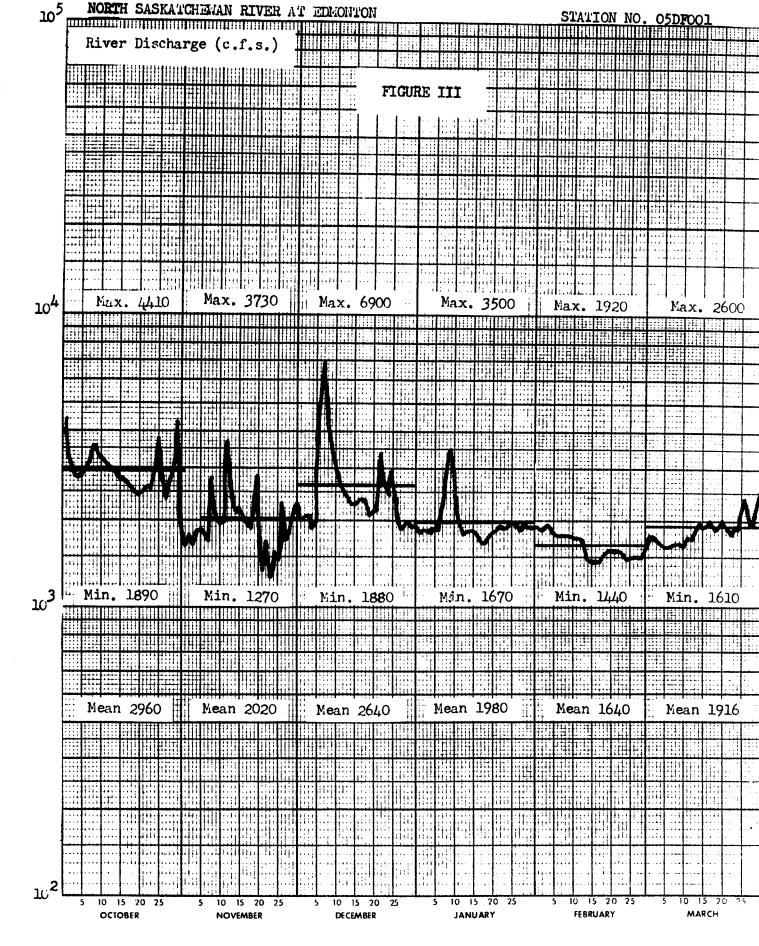
These represented 24-hour composites except for two downstream industries that were 'grab' sampled.

#### NORTH SASKATCHEWAN RIVER FLOWS

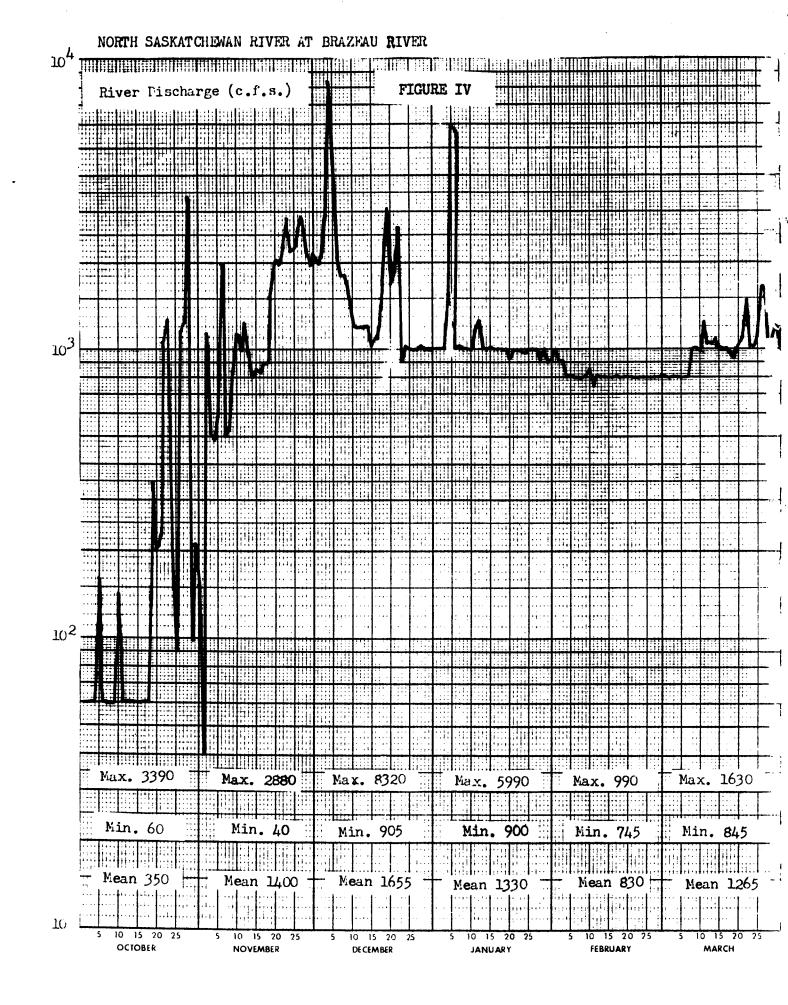
In considering stream pollution, the discharge rate of a river and consequently the dilution factor are of prime importance. Daily discharge in the North Saskatchewan River at Edmonton (Station No. 05DF001) and the discharge rate for the Brazeau Reservoir for the period of October, 1970 through March, 1971 are presented in Figures III and IV.

Ice conditions on the North Saskatchewan River at Edmonton began on October 27, 1970 and remained until April 5, 1971. During this period the minimum monthly mean flow was 1640 c.f.s. recorded in February, while the minimum daily discharge rate was 1270 c.f.s. recorded on November 23, 1970.





ICE CONDITIONS - OCTOBER 27, 1970 to APRIL 5, 1971



Generally flows in the river were considerably less than in previous years, with minimum daily flows below 2000 c.f.s. in every month from October 1970 through March 1971. There were 103 days with flows less than 2000 c.f.s. and 11 days when the flows were 1500 c.f.s. or lower during this five-month period. These low flow conditions are reflected in the frequency distribution of flows, North Saskatchewan River at Edmonton, Figure V.

### LOADINGS TO THE NORTH SASKATCHEMAN RIVER

Table I presents users of the North Saskatchewan River for discharge of waste water. The type of waste and treatment are also included. Najor contributors of wastes to the river were surveyed during the winter. In most cases twenty-four hour composite samples were taken in conjunction with river quality sampling surveys.

Total loadings from 14 major effluents are listed in Table II. Three surveys completed during the summer are included with the nine conducted during the winter months.

Biochemical Oxygen Demand (B.O.D.5) showed loadings varying from 12,200 lbs/day (Narch 9-10, 1971) to 51,700 lbs/day (November 16-19, 1970). Lower loadings for this constituent in 1971 are in part due to a higher degree of treatment of the municipal waste from Edmonton Sewage Treatment Plants.

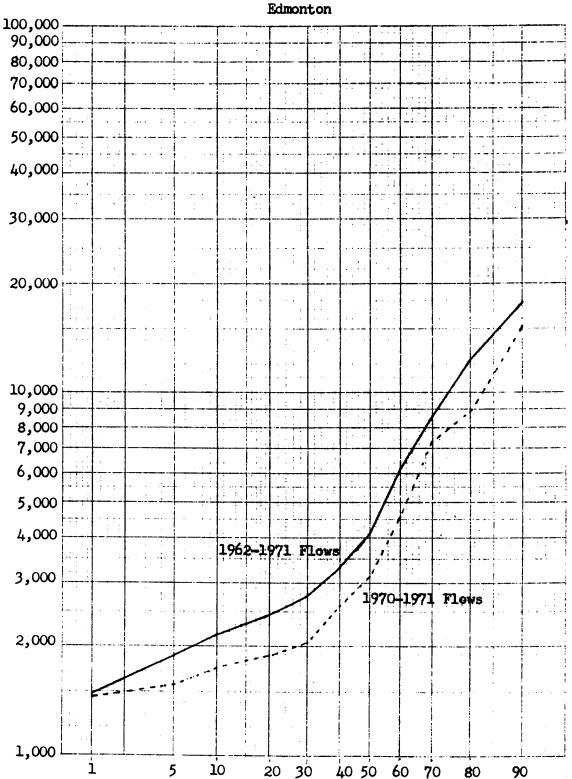
Phenol loadings ranged from 17 lbs/day (January 25-26, 1971) to 68 lbs/day (December 1-2, 1970). Consistently lower loadings were noted for the four survey months of 1971.

Nutrients to the North Saskatchewan River from all sources decreased from loadings of past years. Ammonia Nitrogen and Phosphates loadings decreased significantly.

RIVER DISCHARGE (c.f.s.)

FIGURE V

Statistical Flow Analysis of the North Saskatchewan River at
Edmonton



Frequency of occurrence less than indicated (per cent)

1962 - Mar. 1971 Maximum flow 91,600 cubic feet per second Minimum flow 1,070 cubic feet per second

April 1970 - Mar. 1971 Maximum flow 53,700 cubic feet per second Minimum flow 1,270 cubic feet per second

### TABLE I

### NORTH SASKATCHENAN EFFLUENT DISCHARGERS

Source	Type of Discharge and Treatment	Discharge to
To.m of Rocky Lountain House	Domestic Sewage (L)	N.S.R.
Town of Drayton Valley	Domestic Sewage (L)	N.S.R.
Town of Devon	Domestic Sewage (ST)	N.S.R.
Imperial Gil - Devon	Industrial Waste (L)	N.S.R.
City of Edmonton	Domestic Sewage (ST in Winter)	N.S.R.
Canadian Industries Ltd.	Industrial Waste (WELL)	
Building Products Ltd.	Industrial Waste (AERATED LAGOONS)	N.S.R.
Imperial Gil Ltd.	Industrial Waste (A.P.I., WELL)	N.S.R.
Тежасо	Industrial Waste (A.P.I., WELL, L)	N.S.R.
Union Carbide	Industrial Waste (L)	N.S.R.
Turbo Oil Limited	Industrial Waste (L)	N.S.R.
Gulf Oil Canada Ltd.	Industrial Waste (A.P.I., L)	N.S.R.
Chemcell	Industrial Waste (WELL, L)	N.S.R.
Uniroyal	Industrial Waste (L - SUNMER DISCHARGE)	N.S.R.
Alberta Hospital - Oliver City of Edmonton Packing Plants Sherwood Park	Waste Storage & Disposal during Summer Nonths (L) to City Lagoons	N.S.R.
Town of Fort Saskatchewan	Domestic Sewage (L)	N.S.R.
Sherritt Gordon Mines Ltd.	Industrial Waste (L)	N.S.R.
Dow Chemical Ltd.	Industrial Waste (L, WELL)	N.S.R.
Redwater Imperial Fertilizer	Industrial Waste (L)	N.S.R.
Redwater Imperial Oil Gas Plant	Industrial Waste (L)	N.S.R.
Town of Redwater	Domestic Sewage (L)	N.S.R.
Waskatenau	Domestic Sewage (L)	N.S.R.
Chemcell - Duvernay	Industrial Waste (NT)	N.S.R.
Elk Point	Domestic Sewage (L)	N.S.R.
Canadian Salt Co. Ltd.	Industrial Waste (NT)	N.S.R.
LEGEND: L - Lagoon ST - Secondary Treatmen	NT - No Treatment. AP: - A.P.I. Separator	Q

Ammonia-Nitrogen varied from 9000 lbs/day (November 16-19, 1970) to 18,400 lbs/day (November 2-4, 1970) while Phosphate loadings ranged from 7180 lbs/day (February 20-22, 1971) to 15,400 lbs/day (November 2-4, 1970).

### DISSOLVED OXYGEN IN NORTH SASKATCHEMAN RIVER

River profiles of Dissolved Oxygen are presented in Appendix \*A\*.

These indicate the effect of oxygen demanding material on Dissolved Oxygen from the Brazeau River Discharge to Lloydminster Ferry. The plots show near saturation levels of Dissolved Oxygen for the two surveys conducted in November, 1970. Subsequent to the survey of November 16-19, 1970,

Dissolved Oxygen levels dropped to a low of 5.6 mg/l on February 10, 1971 at Lloydminster Ferry. A reading of 5.7 mg/l was noted during the same survey at Lindbergh and the same reading observed at Lloydminster Ferry during the following survey of February 24, 1971.

Dissolved Oxygen in the river upstream of Edmonton remained above 10 mg/l throughout the winter except for the Brazeau Reservoir discharge location. Here, the oxygen level dropped to a low of 4.8 mg/l on January 12, 1971. Other readings subsequent to that date ranged from 8.4 mg/l to 8.9 mg/l.

#### BIOCHEMICAL OXYGEN DEMAND

The amount of oxygen consumed in stabilizing waste material is determined by the Biochemical Oxygen Demand test. These tests were run on all river samples taken during the sampling period.

Background levels of Biochemical Oxygen Demand ranged from 0.1 mg/1 to 3.4 mg/l between the Brazeau Reservoir and Edmonton. Concentrations at 105 Street Bridge varied from 0.1 mg/l (November 4, 1970) to 1.3 mg/l (Earch 10, 1971).

# TABLE II-a NORTH SASKATCHEJAH RIVER

# Total Loadings - 1bs/day

Sumpling Date:	<b>Junc</b> 16 <b>–17/7</b> 0	July 13 <b>–</b> 15/70	September 22 <b>–</b> 23/70	November .2 <b>-</b> /4/70
River Flow at Edmonton (c.f.s)	18,900	14,000	2,730	1,740
Total Number of Locations: Number of Locations Sampled:	14 12	14,	1 <i>1</i> , 13	14,
Biochemical Organ Demand	78,900	14,500	92,500	49,900
Chemical Orggen Demand	204,000	171,000	242,000	152,000
P.P. Alkalinity as CaCO3	9,2%	93,400	38,500	596
Total Alkalinity " "	155,000	238,000	172,000	167,666
Total Residue	962,000	843,CCC	<b>7</b> 08,000	658,000
Ignition Loss	140,000	195,000	267,000	221,000
Non-filtrable Residue	133,000	8 <b>7,2</b> 00	196,000	66,500
Ignition Loss	64,100	26,000	89,900	15,400
Oil and Grease	19,100	14,400	30,50C	2,150
Phonols	115	87.2	29.1	20.4
Chlorides	387,000	197,000	117,000	IJ₄ℰ <b>,</b> ССС
Total Phosphorous as Po	6,530	6,6%	o <b>,</b> 740	15,400
Amonia Nitrogen	7,690	14,800	15,700	18,374
Nitrate Nitrogen	1,030	189	<b>25</b> 9	2,41(
Sulphates	8,460	<b>39,1</b> 00	10,700	20,900
Hexm.valent Chromium	11.0	9.0	13.0	12.0
Fluorides	164	æ.3	50.9	31.7
Coppor	1.1	0.0	C.2	C.7
Nickel	1.6	0.9	0.0	c.o
Cynnide	6.7	6.7	1.8	C.7
Arsenic	-	3.3	-	-

TABLE II-b

NORTH SASKATCHE AN RIVER (Cont.)

## Total Londings - 1bs/day

Sampling Date:	Hovember 16 <b>–</b> 19/70	December 1-2/70	December 14-16/70	January 10 <b>–1</b> 3/71
River Flow at Edmonton (c.f.s.)	1,990	2,030	2,310	1,830
Total Number of Locations: Number of Locations Sampled:	υ, υ,	14 14	カ* カ*	14,
Biochemical Chygen Demnid	51,700	28,100	49,766	<b>26,3</b> 00
Chemical Crygen Demand	82,100	80,700	116,000	<b>55,9</b> 00
P.P. Alkalinity as and3	96,200	12,800	24,700	46,200
Total Alkalinity " "	201,000	135,500	132,000	159,000
Total Residue	568,000	491,000	856,000	487,000
Ignition Loss	147,000	59,800	354 <b>,</b> 000	94,300
Non-filtrable Residue	66,600	52,400	43,700	36,500
Imition Loss	32,500	26,400	23,700	21,900
Gil and Grease	6,940	1,151	2,546	4,590
Phenols	33.6	6 <b>7.</b> 6	18.3	<b>23.</b> 8
Chlorides	105,000	97,500	164,20	110,000
Total Phosphorous as PO#	7,930	9,210	8,430	e,500
Almonia Nitrogen	9,000	13,160	15,700	10,500
Mitrate Mitrogen	.226	49	113	873
Sulphates	12,900	24,20	16,200	15,600
Homevelent Chromium	<b>36.</b> 0	7.0	<b>3</b> C.C	35.0
Fluorides	4.9	34.8	17.7	9.1
Copper	2.5	0.6	C.5	<b>c.</b> c
Nickel	1,.0	6.0	6.7	9.4
Cynnide	<b>6.</b> 0	U.0	٥.6	٥.،٥
Arsonic	-		-	

## NORTH SASKATCHEWAN RIVER (Cont d.)

# Total Loadings - lbs/day

Sampling Date:	January 25 <b>–</b> 26/71	February 8–10/71	February 20–22/71	larch 9-10/71
River Flow at Edmonton (c.f.s.):	1,975	1,770	1,510	1,650
Total Number of Locations: Number of Locations Sampled:	14 14	14 14	14 14	14 13
Biochemical Oxygen Demand	30,400	26,800	18,580	12,200
Chemical Oxygen Demand	103,000	99 <b>,7</b> 00	84,900	76,500
P.P. Alkalinity as CaO3	42,700	54,300	36,900	4,260
Total Alkalinity " "	142,000	146,000	170,000	121,000
Total Residue	1,780,000	1,460,000	891,000	1,700,000
Ignition Loss	117,000	370,000	121,000	143,000
Non-filtrable Residue	56,100	<i>3</i> 0 <b>,7</b> 00	68,800	48,200
Ignition Loss	24,800	10,700	25,000	21,400
Oil and Grease	4,920	5,070	4,670	6,000
Phenols	17.3	21.0	23.1	24.8
Chlorides	786,000	555,000	234,000	449 <b>,</b> 000
Total Phosphorous as PO\=	8,960	7,510	7,180	8,270
Ammonia Nitrogen	10,300	9,630	10,400	13,600
Nitrate Nitrogen	341	689	875	<b>7</b> 69
Sulphates	30,300	20,100	104,000	40,900
Hexavalent Chromium	50.0	23.0	45.0	<b>7</b> 0.0
Fluorides	10.7	20.3	26.3	31.8
Copper	2.1	1.6	4.3	4.0
Nickel	10.1	6.5	29.4	540
Cyanide	0.0	0.6	0.0	0.0
Arsenic		-		<del></del>

The highest values for Biochemical Oxygen Demand downstream of Edmonton were observed in the zones of degradation and decomposition, as expected. At Fort Saskatchewan Bridge (zone of degradation) values ranged from 1.3 mg/l to 5.2 mg/l, with mean and median values of 2.9 mg/l and 2.5 mg/l, respectively. Concentrations at Vinca Bridge and Waskatenau Bridge (zone of decomposition) were only slightly lower.

A decrease is noted in Biochemical Oxygen Demand from Duvernay
Bridge to Lloydminster Ferry. The highest value in this region was
3.9 mg/l at both Lindbergh and Lloydminster Ferry on November 4, 1970.
The lowest value was 0.4 mg/l on December 2, 1970 at Lloydminster Ferry.

#### NUTRIENTS

Nutrient concentrations of Ammonia Nitrogen, Nitrate Nitrogen, and Total Phosphorous remained relatively constant above Edmonton. Minimum and maximum values for these three constituents were 0.0 to 0.6 mg/l of Ammonia Nitrogen, 0.0 to 0.5 mg/l of Nitrate Nitrogen, and 0.0 to 0.7 mg/l of Total Phosphorous.

The effects of both municipal and industrial loadings are reflected on these constituents downstream from Edmonton. Maximum values observed were 2.4 mg/l for Ammonia Nitrogen, 1.1 mg/l for Nitrate Nitrogen and 15 mg/l for Total Phosphorous.

Total Phosphorous above 2.5 mg/l are not consistent with results obtained in other surveys of the river. This indicates either that a sample of a slug flow high in phosphates was taken during a survey or that the sample was contaminated.

#### PHENOLS

Samples taken at four locations upstream of Edmonton showed a

relatively high concentration of Phenols, especially those taken in November and December. The maximum value was 15 p.p.b. on the sample of November 18, 1971 taken at 105 Street Bridge, Edmonton. A greater portion of the analysis (29 of 36 samples) showed values within the established criterion level of 5 p.p.b.

The highest value for Phenols downstream of Edmonton was 14 p.p.b. This value was determined on the sample taken at the Fort Saskatchewan Bridge on November 18, 1970.

Fifty-four individual samples were collected from the six locations downstream of Edmonton during the winter sampling period (November 1, 1970 to March 31, 1971). Twenty of these samples contained Phenols exceeding 5 p.p.b. Waskatenau Bridge and Duvernay Bridge locations had the highest incidence of Phenols exceeding the established criterion level.

#### THRESHOLD ODOR NUMBER

Threshold Odor of the river water from the Brazeau Dam to Edmonton was of the "Musty" type with a maximum value of 8. All other values were between 1 and 4 with one incident of a "Chemical Hydrocarbon" type determined on a sample taken at Devon Bridge on February 9, 1971.

Odors were of a higher magnitude downstream of Edmonton, especially at Waskatenau Bridge where a value of 32 "Chemical" was determined on the sample of December 17, 1970.

There were no complaints of odors in the river water this past winter from downstream users.

#### OIL AND GREASE

Oil and Grease concentrations in the North Saskatchewan River during the winter sampling program were between 0.3 mg/l to 4.6 mg/l. The higher

values may have been due to unauthorized release of oily matter to the North Saskatchewan River. Generally, the concentration was the same or slightly improved from conditions experienced last year.

#### HEAVY METALS

One complete survey of the North Saskatchewan River was made during February, 1971 to determine the concentration of heavy metals. The analytical results for these tests are presented in Table III-a. These results show no appreciable change in concentration of metals from the Brazeau Dam to Lloydminster Ferry except for Chromium and Mercury.

The concentration of Chromium showed a general upward trend starting from the Brazeau Dam (2 p.p.b.) to Vinca Bridge (17.5 p.p.b.). The concentration then decreased to 5 p.p.b. at Lloydminster Ferry.

Samples for Mercury were taken at Edmonton's 105 Street Bridge,
Vinca Bridge, and Lloydminster Ferry. Increases are noted in both Soluble
and Total Mercury concentrations, progressing downstream to the AlbertaSaskatchewan border. The values ranged from 0.08 p.p.b. to 0.16 p.p.b.
for Soluble Mercury and 0.11 p.p.b. to 0.43 p.p.b. for Total Mercury.

Table III-b presents heavy metal results of samples taken at Edmonton's 105 Street Bridge and Vinca Bridge in May, 1971. Concentrations of Selenium and Mercury were above the established Water Quality Criteria level.

#### PESTICIDES

Samples for Pesticides analyses were taken during June and July, 1970 and April and May, 1971, from various locations on the North Saskatchewan River and Whitemud Creek. Samples were also obtained from three storm sewer outlets in Edmonton on April 1, 1971.

TABLE III a

HEAVY METALS

(Concentration in Parts Per Billion)

Sampling Location:	.ocation:	Brazeau Dam	Drayton Valley	Devon	105 St. Bridge	Ft. Sask.	Vinca	Waska- tenau	Duver- nay	Lind- bergh	Lloyd- minste
Sampling Date	ate :	2/23/71	2/23/71	2/23/71	2/23/71	2/24/71	2/24/71	2/24/71	2/24/71	2/57/71	2/25/7
Cadmium	- Total	0	2.0	1.0	1.5	2.0	2.0	2.0	0.4	2.5	1.5
Manganese - Total	- Total	17	50	15	7.5	32	18.5	19	15	7.5	7.5
Zinc	- Total	7.5	10	10	7.5	15	12.5	22.5	ដ	12.5	7.5
Copper	- Total	7.5	10	10	15	11.5	6	Ħ	5.0	5.0	15
Cobalt	- Total	2.5	2.0	3.5	3.0	6.5	6.5	5.0	O	0	7.0
Nickel	- Totel	12.5	12.5	15	15	OJ.	12.5	12.5	20	15	17.5
Chronium	- Total	<b>%</b>	3.5	5.0	2.5	5.0	17.5	7.5	1.5	01	5.0
Lead	- Totel	42.5	25	27.5	25	42.5	37.5	27.5	27.5	15	22.5
Tin	- Total	50	25	25	25	25	90	25	55	35	25
Arsenic	- Total	i	ı	1	1	ı	ı	ı	ı	i	ŀ
Selenium	- Totel	1	ı	t	ı	ı	ı	ı	ı	ı	ı
hercury	- Soluble - Total	1 1	1 1	1 1	8.i.	1 1	.10	1 1	1 1	1 1	.16

## TABLE III-b

### HEAVY METALS

## NORTH SASKATCHEWAN RIVER

(Concentration in Parts Per Billion)

Sampling L	ocation :	105 St. Bridge	Vinca
Sampling D	ate :	May 21, 1971	May 21, 1971
CADMIUM	- Total	2	2
	- Dissolved	2	2
CHROMIUM	- Total	5	10
	- Dissolved	5	10
COBALT	- Total	5	0
	- Dissolved	5	0
COPPER	- Total	5	8
	- Dissolved	5	8
LEAD	- Total - Dissolved	8	15 15
MANGANESE	- Total	18	48
	- Dissolved	5	48
MERCURY	- Total	0.97	0.58
	- Dissolved	0.14	0.53
NICKEL	- Total	8	10
	- Dissolved	<del>-</del>	10
SELENIUM	- Total	200	. 15
	- Dissolved	-	15
TIN	- Total	<b>7</b> 0	17
	- Dissolved	<b>7</b> 0	17
ZINC	- Total	5	25
	- Dissolved	5	8

Thirty compounds were tested on each sample taken in 1970 and forty compounds were tested on each sample taken in 1971. Pesticide compounds identified in the samples include 2,4-D Acid, 2,4,5-T Acid, M.C.P.A. Acid, Dicamba, Phorate, and Diazinon. Other compounds that were detected but could not be identified are reported in a general classification of pesticides as: Acids, Chlorinated, Phosphorous and Esters (See Appendix \*C\*).

2,4-D Acid was detected in six samples. The maximum value was 0.4 p.p.b. on the sample taken at Fort Saskatchewan Bridge on June 30, 1970.

2,4,5-T Acid was detected in three samples. The maximum concentration was 0.1 p.p.b. on the sample taken at Fort Saskatchewan Bridge on June 30, 1970.

M.C.P.A. Acid was detected in five samples. The maximum concentration was 0.6 p.p.b. on samples taken at Vinca Bridge on June 23, 1970. Trace quantities of two other pesticides were detected in storm sewer samples taken on April 1, 1971. Diazinon was determined on a sample from a storm sewer in Emily Murphy Park and Phorate was found in a sample from a storm sewer near Whitemud Road.

#### BACTERIOLOGICAL CONSTITUENTS

Coliform, <u>E. coli</u> and Standard Plate Count were determined on all samples taken from the North Saskatchewan River during the winter survey. A gradual increase in these constituents was observed from the Brazeau Reservoir to Edmonton. Maximum values obtained in this stretch of the river were 2400+ organisms/100 ml for Coliform, 170 organisms/100 ml for E. coli and 55,000 colonies/ml for Standard Plate Count.

Downstream from Edmonton at Ft. Saskatchewan Bridge, all bacterio-logical constituents showed sharp increases as expected. Maximum values were 180,000 organisms/100 ml for Coliform, 2700 organisms/100 ml for E. coli and 1,100,000 colonies/ml for Standard Plant Count. Values determined at Vinca Bridge was similar to those found at Ft. Saskatchewan Bridge.

A decrease in bacteriological constituents is noted as one proceeds further downstream. Values found at Lloydminster Ferry location were similar to those determined at Edmonton's 105 Street Bridge location.

#### DISCUSSION

A comparison of frequency distribution of flows shows a lower rate of discharge of water in the North Saskatchewan River this past winter than in the previous ten years. This is reflected on the quality of the river water downstream from Edmonton.

Lower Dissolved Oxygen and generally higher concentrations of Biochemical Oxygen Demand, Alkalinity, Total Residue, Chlorides, Nutrients, and Phenolics were noted as compared to data obtained in the previous year.

Levels of Ammonia Nitrogen and Total Phosphorous continue to be excessive despite a significant decrease in loadings of these constituents.

Bacteriological constituents downstream of Edmonton were in excess of established objectives for Total Coliform and E. coli.

Concentrations of Heavy Metals did not vary appreciably from background levels except for Chromium and Mercury. Further investigations are required to substantiate and verify the concentrations of Total Mercury determined at Vinca Bridge and Lloydminster Ferry.

#### 700017117

### WATER POLLUTION CONTROL TECHNOLOGY

This is a selected list containing the key terms likely to be encountered in water pollution control technology.

- ABSORPTION The taking up of one substance into the body of another.
- ACID Most commonly refers to a large class of chemicals having a sour taste in water, ability to dissolve certain metals, bases or alkalies to form salts and to turn certain acid-base indicators to their acid form. Characterized by the hydrated H<sup>+</sup> ion.
- ADSORPTION The taking up of one substance upon the surface of or interface zone of another substance.
- ADVANCED WASTE TREATHENT Renovation of used water by biological, chemical or physical methods that are applied to upgrade water quality for specific reuse requirements. May include more efficient cleanup of a general nature or the removal of components that are inefficiently removed by conventional treatment processes.
- AERATION The operation of adding oxygen to, removing volatile constituents from, or mixing a liquid by intimate contact with air.
- AEROBIC A condition characterized by an excess of dissolved oxygen in the aquatic environment.
- AEROBIC BACTERIA Organisms that require dissolved oxygen in the aquatic environment to enable them to metabolize or to grow.
- AIGAE Primitive plants, one or many celled, usually aquatic and capable of growth on mineral materials via energy from the sun and the green coloring material, chlorophyll. Generally considered as the primary source of food for all other organisms.
- ALKALINITY A term used to represent the sum of the effects opposite in reaction to acids in water. Usually due to carbonates, bicarbonates and hydroxides; also including borates, silicates and phosphates.
- ANAEROBIC A condition in which dissolved oxygen is not detectable in the aquatic environment. Commonly characterized by the formation of reduced sulfur compounds from the use of bound oxygen from sulfates as an hydrogen acceptor.
- ANAEROBIC BACTERIA Organisms that can metabolize and grow in the absence of dissolved oxygen. Their oxygen supply is obtained from the bound oxygen such as in sulfates, carbonates, or other oxygen-containing compounds.
- ANION A negatively charged ion in water solution. May be a single element or a combination of elements, such as the Cl ion in a water solution of NaCl (common table salt) or SOL ion in a sulfuric acid solution.

- ASSESSMENT A legal financial obligation of the property owner in an irrigation, water, drainage or sanitary district, created for the purpose of financing the construction and operation of facilities required to protect and enhance public benefit within the district.
- BACTERIA Primitive organisms having some of the features of plants and animals. Generally included among the fungi. Usually do not contain chlorophyll, hence commonly require preformed organic nutrients among their foods. May exist as single cells, groups, filaments, or colonies.
- BENTHIC DEPOSIT (BENTHOS) Refers to the accumulated deposition of cell mass living or dead that collects at the bottom of a stream impoundment where velocity or catchment permits.
- BIO-CHETCAL Resulting from the combined activities of biological and chemical transformations. Usually measured in terms of the ensuing chemical changes.
- BIODEGRADATION The stabilization of wastewater contaminants by biological conversion of pollutants into separatable materials at a higher oxidation state.
- BIOLOGICAL PROCESSES Activities of living organisms to sustain life, growth, and reproduction. Commonly the processes by which organisms degrade complex organic material into simpler substances at a higher oxidation state to obtain energy for life processes and growth of new cell mass.
- BOD Biological or biochemical oxygen demand. A test for estimation of wastewater polluting effects in terms of the oxygen requirements for biochemical stabilization under specified conditions and time.
- BUFFER ACTION An action exhibited by certain chemicals that limits the change in pH upon addition of acid or alkaline materials to the system. In surface water, the primary buffer action is related to carbon dioxide, bicarbonate and carbonate equilibria.
- CARBOHYDRATES Naturally occurring compounds consisting of carbon, hydrogen and oxygen, that are considered as energy foods and precursors of proteins and fats in the natural food chain.
- CATION A positively charged ion in water solution. May be a single element or a combination of elements, such as Na<sup>+</sup> in a water solution of NaCl (common table salt).
- CHIOROPHYLL The green coloring material or pigments in plants that promotes the photosynthetic reactions forming organic materials from inorganic nutrients and light energy within the living cells.
- CIARIFTER A basin or chamber serving as an enlargement of a channel to reduce flow velocity sufficiently to permit separation of settleable or floatable materials from the carrier water (a sedimentation basin).
- COAGULANT A chemical, or chemicals, which when added to water suspensions will cause finely dispersed materials to gather into larger masses of improved filterability, settleability, or drainability.

- COAGULATION The process of modifying chemical, physical, or biological conditions to cause flocculation or agglomeration of particulates.
- COD A test for the estimation of the contamination of a wastewater in terms of oxygen requirements from a strong chemical oxidant under specified conditions, i.e., Dichromate, 50% sulfuric acid and 145°C for 2 hours.
- COLIFORI GROUP A group of bacteria that inhabits the intestinal tract of man, warm-blooded animals, and may be found in plants, soil, air and the aquatic environment. Includes aerobic and facultative gram negative non-spore forming bacilli that ferment lactose with gas formation.
- COLLOID, COLLOIDAL STATE A state of suspension in which the particulate or insoluble material is in a finely divided form that remains dispersed in the liquid for extended time periods. Usually cloudy or turbid suspensions requiring flocculation before clarification.

#### CONCENTRATION -

- a) The act of increasing the mass per unit volume of one substance with respect to another, such as concentrating the solids in a sludge from 3% to 6%.
- b) A means of designating the ratio of one substance with respect to another, such as 15 mg of suspended solids per liter of water.
- CRITERION (pl. CRITERIA) Something which can be measured. Commonly used as a basis for standards.
- CUBIC FOOT PER SECOND (c.f.s.) A unit of discharge rate such as one cubic foot of water per second past a given point.
- DATA Records of observations or measurements of facts, occurrences and conditions in written, graphical or tabular form.
- DEGRADE To reduce the complexity of a chemical compound.

#### DENITRIFICATION -

- a) The conversion of oxidized nitrogen (nitrate and nitrite-N) to nitrogen gas by contact with septic wastewater solids or other reducing chemicals.
- b) A reduction process with respect to oxidized nitrogen.
- DETERGENT Something used for cleaning. Commonly consists of soap or surfactant plus various additives or associated materials.

#### DILUTION -

- a) To make thinner or more liquid.
- b) A ratio, volume or weight of a more concentrated sample or effluent flow compared to that into which it is discharged.

#### DISSOLVED -

- a) Those materials dispersed in water in ionic, atomic, or molecular form; an homogenous mixture or solution.
- b) Generally clear but may be colored.
- c) Present in true solution form.

- DISSOLVED OXYGEN (D.O.) Dissolved molecular oxygen usually expressed in mg DO/1 or percent of saturation.
- ECOLOGY The relation of an organism to its environment; i.e., how is an organism affected by its surroundings such as air, water, heat, noise, contamination, etc.
- EFFLUINT A liquid or product water discharged from a chamber, basin or other treatment operation.
- ENTERIC ORGANISMS Those organisms commonly associated with the intestinal tract.
- EMTRAINEMT A condition or action that will cause an immiscible substance to be mixed with another. Usually the result of turbulence or entrapment; i.e., air bubbles in aqueous media.
- ENZYME A soluble or colloidal organic catalyst produced by a living organism. Usually they are simple or conjugated proteins that catalyze specific reactions.
- EUTROPHIC Well nourished; rich in dissolved nutrients.
- EUTROPHICATION An action involving the aging of lakes characterized by nutrient enrichment and increasing growth of plant and animal organisms. The net effect is to decrease depth until the lake becomes a bog and eventually dry land. Man-made pollution tends to hasten the process.
- FACULTATIVE BACTERIA Bacteria that can adapt themselves to growth and metabolism under aerobic or anaerobic conditions. Many organisms of interest in wastewater stabilization are among this group.
- FIOC Gelatinous or amorphous solids formed by chemical, biological or physical agglomeration of fine materials into larger masses that are more readily separated from the liquid.
- FUNGI Simple or complex organisms without chlorophyll. The simpler forms are one-celled; higher forms have branched filaments and complicated life cycles. Examples are molds, yeasts and mushrooms.
- HARDNESS Commonly refers to the chemicals interfering with soap action or producing scale in boilers or heating units. Specifically refers to Calcium and Magnesium salts; sometimes including iron, aluminum, and silica.
- INDICATOR May include the color change of a dye, electronic sensor response, or other means of estimating the equivalence point of a reaction between two different materials.
- INFLUENT That material entering a process unit or operation.
- INORGANIC Being composed of material other than plant or animal materials. Forming or belonging to the inanimate world.
- LAGOON A natural or artificial basin used for storage and/or stabilization of wastewater or sludge. Sometimes used for indefinite storage for disposal purposes. Commonly the lagoon depth is greater than a wadable depth but not greater than twenty feet.

- IOAD The load to a process is that which is contained in the inflow to that process. It may be expressed as hydraulic, oxygen demand, solids, or other criteria.
- MICRO ORGANISM Commonly an organism too small to be observed individually by the human eye without optical aid.
- NG/L A unit of concentration on a weight/volume basis: hilligrams per liter. Equivalent to ppm when the specific gravity of the liquid is 1.0.
- MIXING ZONE An area where two or more substances of different characteristics blend to form a uniform mixture; i.e., chlorine application, heated water, or other discharged materials entering a water mass will show significant differences of chlorine residual, temperature or other criteria, depending upon the sampling location throughout the mixing zone and approach uniform results with respect to lateral, longitudinal, and vertical sampling positions when mixing has been completed.
- MOST PROBABLE NUMBER (MPN) A statistical method of determining microbial populations. A multiple dilution tube technique is utilized with a standard medium and observations are made for specific individual tube effects. Resultant coding is translated by mathematical probability tables into population numbers.
- NITRIFICATION The biochemical conversion of unoxidized nitrogen (ammonia and organic N) to oxidized nitrogen (usually nitrate).

#### NUTRIENTS -

- a) Anything essential to support life.
- b) Includes many common elements and combinations of them. The major nutrients include carbon, hydrogen, oxygen, nitrogen, sulfur, and phosphorus.
- c) Nitrogen and phosphorus are of major concern because they tend to recycle and are hard to control.

#### OILS -

- a) Liquid fats of animal or vegetable origin.
- b) Oily or waxy mineral oils.
- OXIDATION POND -A shallow basin employed for the stabilization of wastewaters.
- OXYGEN DEPLETION The loss of oxygen from water or sewage due to biological, chemical or physical action.
- PARASITE A living organism deriving its nutrients at the expense of another living organism, giving nothing in return.
- PARSHALL FIGURE A device for estimation of the flow in an open conduit.

  Consists of a constricting section, a throat, and an expanding section.

  The throat contains a sill over which the liquid passes. The pressure change over the sill can be related to quantity of flow.

- PARTS PER MILLION (PPM) A unit of concentration signifying perts of some substance per million parts of dispersing medium. Equivalent numerically to mg/l only when the specific gravity of the solution is 1.0.
- PATHOGENIC ORGANISMS Bacterial, fungal, viral, or other organisms directly involved with diseases of plants, animals, or man, are included among this group.
- pH An index of hydrogen ion activity. Defined as the negative logarithm (base 10) of H<sup>+</sup> ion concentration at a given instant. On a scale of 0 to 14 pH 7.0 is neutral, pH less than 7.0 indicates a predominance of H<sup>+</sup> or acid ions; pH greater than 7.0 indicates a predominance of OH<sup>-</sup> or alkaline ions.
- POLLUTION Anything appearing in water that renders it unacceptable in terms of established water quality standards. Commonly conditions or contaminants that interfere with subsequent beneficial uses of the water.
- PRIMARY TREATHANT Commonly the separation of settleable or floatable materials from carrier water. Usually preceded by pretreatment such as coarse screens, grit separation, comminution.
- SATURATION Commonly refers to the maximum amount of any material that can be dissolved in water or other liquid at a given temperature and pressure. For oxygen, this commonly refers to a percentage saturation in terms of the saturation value, such as about 9 mg 02/1 at 2000.
- SECOND FOOT An abbreviation for cubic foot per second. A rate term.
- SECONDARY TREATMENT Processes used to convert dissolved and colloidal materials in wastewater to a form that may be separated from the water. Commonly consists of biodegradation and conversion to cell mass in a separatable form with partial oxidation, such as in activated sludge, trickling filtration, or oxidation ponds.

#### SETTLEABLE SOLIDS -

- a) Includes materials that will settle by gravity under low flow velocities.
- b) Commonly expressed in terms of the volume of solids accumulating in an Imhoff cone after one hour on a volume basis.
- SEMER A pipe or conduit, generally covered, for the purposes of conveying wastewaters from the point of origin to a point of treatment or discharge.
- SEMERAGE SYSTEM A system of sewers and appurtenances for the collection, transportation and pumping of used waters for a given area or basin. Any treatment device or facility and its outfall conduit are a part of the system.
- SLIME SEMAGE SLIMES Consisting of organisms growing on wastewater nutrients with the formation of mucilaginous covering, streamers or clumps. May consist of bacteria, molds, protozoa or algae.

#### STABILIZATION -

- a) The activity proceeding along the pathway to stability.
- b) In organic wastes, generally refers to oxidation via biochemical pathways and conversion to gaseous or insoluble materials relatively inert to further change.
- STANDARD Something set by authority. Having qualities or attributes required by law and defined by minimum or maximum limits of acceptability in terms of established criteria or measurable indices.
- STANDARD METHODS Nethods of analysis prescribed by joint action of APHA, ASCE, ANNA, and FWPCA. Nethods accepted by authority.

#### SUBSTRATE -

- a) The base or media in which an organism lives.
- b) The liquid in an activated sludge aeration tank.

#### SURFACTANT -

- a) A chemical that, when added to water, will greatly reduce the surface tension of the solution.
- b) The surface active component in a detergent mixture.
- SUSPENDED SOLIDS The concentration of insoluble materials suspended or dispersed in waste or used water. Generally expressed in mg/liter on a dry weight basis. Usually determined by filtration methods.
- SYMERGISM Refers to the action produced when two or more substances in combination have a greater effect than that produced by the additive effects of each one separately.
- THERIAL POLLUTION Refers to heated discharges to surface waterways. The largest contributor of heated discharges is associated with power production.
- TOC Total Organic Carbon. A test expressing wastewater contaminant concentration in terms of the carbon content.
- TOTAL SOLIDS Refers to the solids contained in dissolved and suspended form in water. Commonly determined on a weight basis by evaporation to dryness.
- VELOCITY (FLOW) A rate term expressed in terms of linear movement per unit of time. Commonly expressed in ft per sec (English) or cm/sec (Metric).

#### VOLATILE MATERIAL -

- a) Refers to those chemicals having a vapor pressure low enough to evaporate from water readily at normal temperatures.
- b) With reference to dry solids, the term includes loss in weight upon ignition at 600°C.

- WATER QUALITY CRITERIA Includes selected analytical measurements with limits designated to be acceptable or unacceptable in reference to water quality standards.
- WATER QUALITY STANDARDS Limits set by authority on the basis of water quality criteria required for beneficial uses.
- WASTEMATER Refers to the used water of a community. Generally contaminated by the waste products from household, commercial or industrial activities. Often contains surface wash, storm water and infiltrations water.

## **ABBREVIATIONS**

## ODOR TYPE

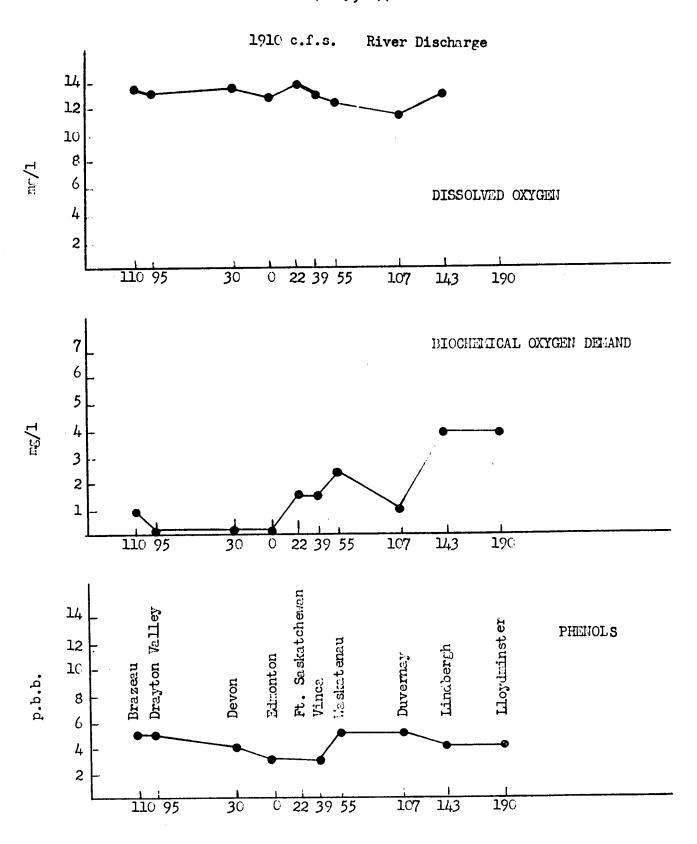
Abbreviation	Nature of Odor	Description (Such as Odors of:)
A AC	Aromatic (spicy) cucumber	Camphor, cloves, lavender, lemon Synura
B BG BN BS BV	Balsamic (flowery) geranium nasturtium sweetish violets	Geranium, violet, vanilla Asterionella Aphanizomenon Coelosphaerium Mallomonas
C CC CH CM CS	Chemical chlorinous hydrocarbons medicinal sulfuretted	Industrial wastes or treatment chemical Free Chlorine Oil Refinery Wastes Phenol and Iodoforms Hydrogen Sulphide
D DF DP DS	Disagreeable fishy pigpen septic	(Pronounced unpleasant odors) Uroglenopsis and Dinobryon Anabaena Stale sewage
E EP	Earthy peaty	Damp earth Peat
G	Grassy	Crushed grass
M MN	Musty moldy	Decomposing straw Damp Cellar
<b>v</b>	Vegetable	Root vegetables
WR	Wood Resin	

# APPENDIX A

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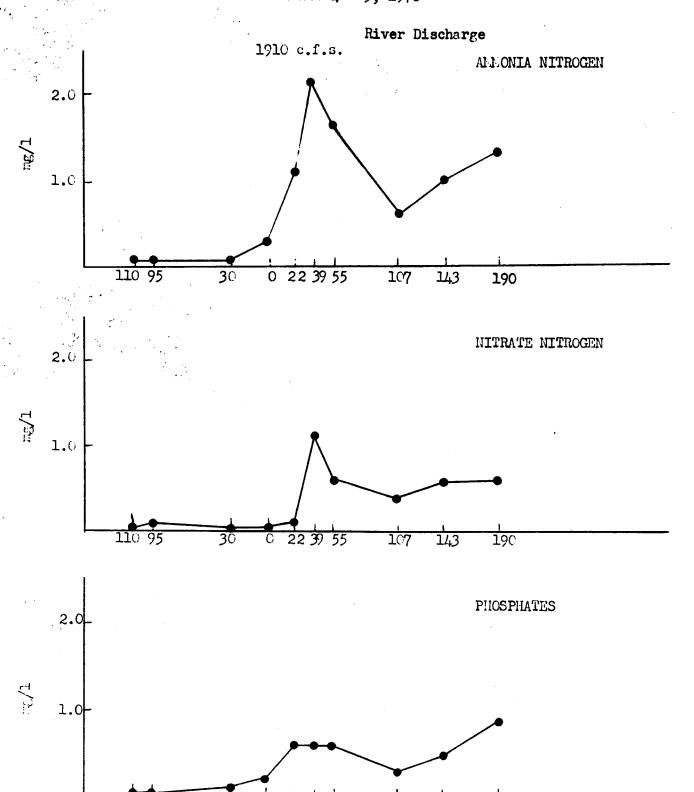
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#### NORTH SASKATCHEWAN RIVER SAMPLING RESULTS November 4 - 5, 1970



RIVER MILES FROM EDMONTON

### NORTH SASKATCHEVAN RIVER SAMPLING RESULTS November 4 - 5, 1970



RIVER NILES FROM EDMONTON

107

143

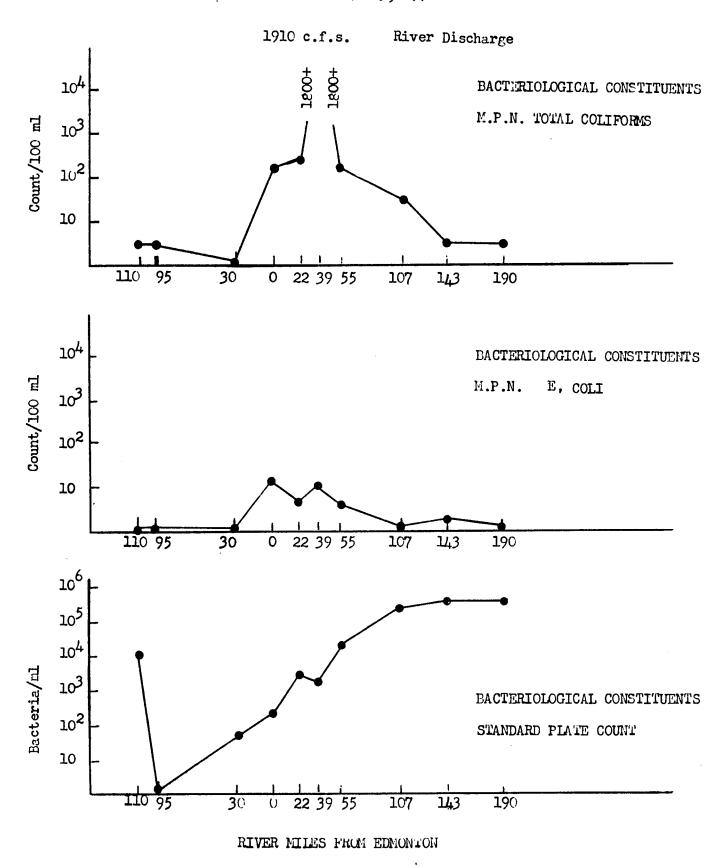
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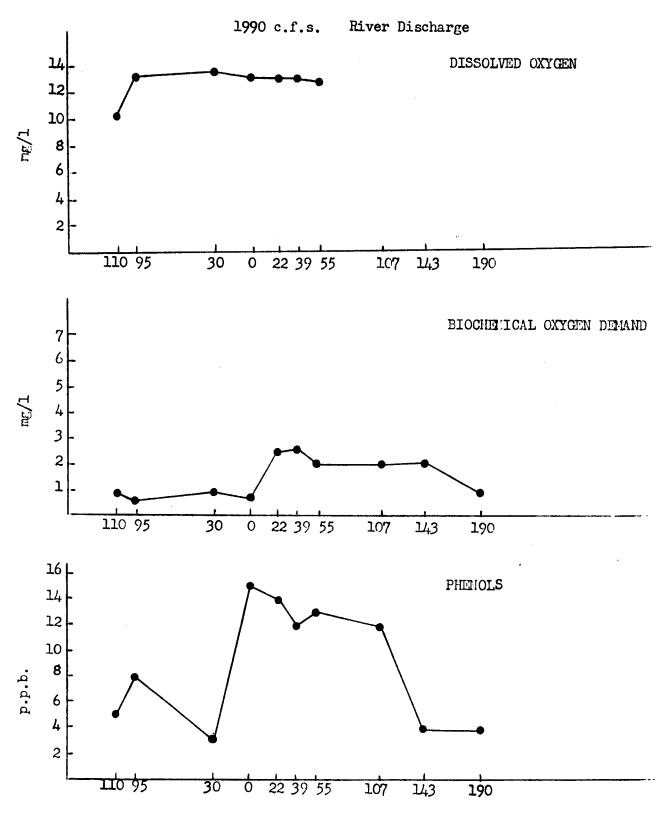
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#### NORTH SASKATCHEWAN RIVER SAMPLING RESULTS November 4 - 5, 1970

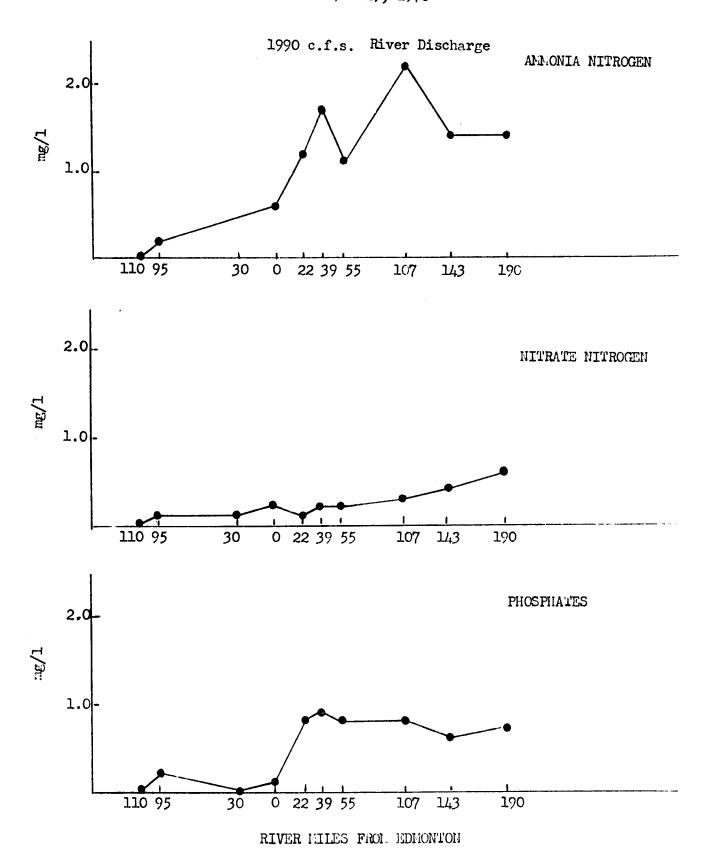


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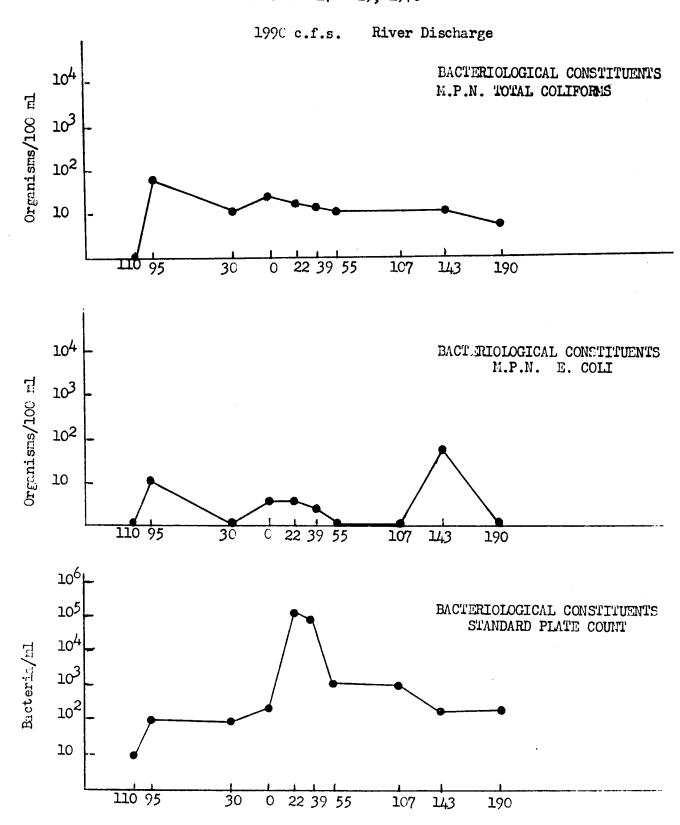


RIVER LILES FROM EDMONTON

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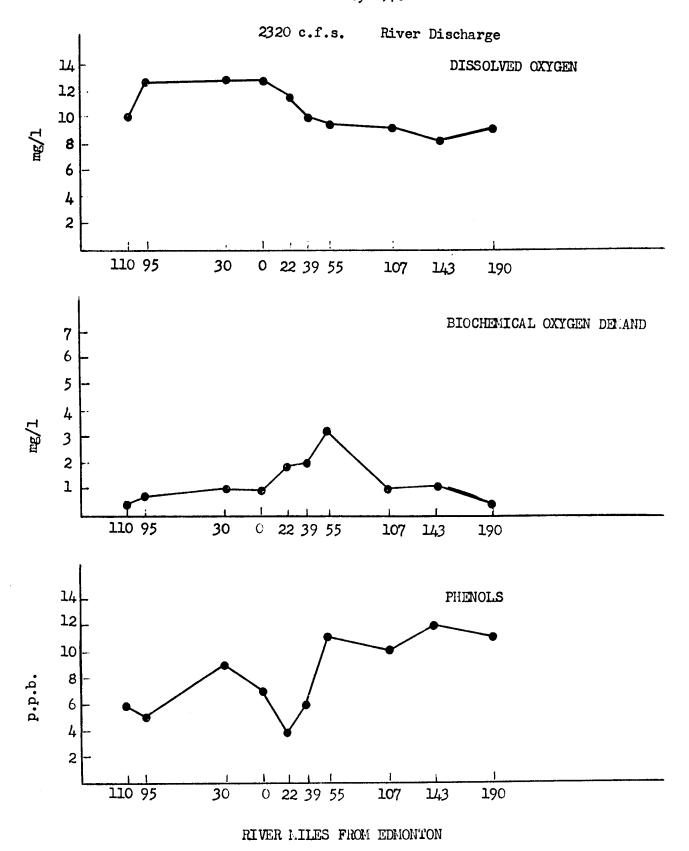


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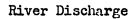


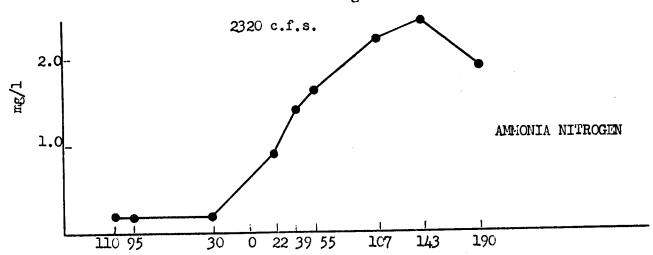
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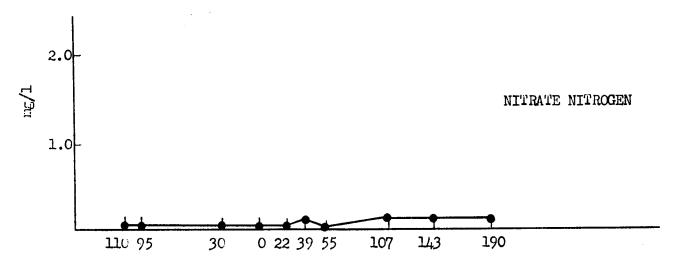
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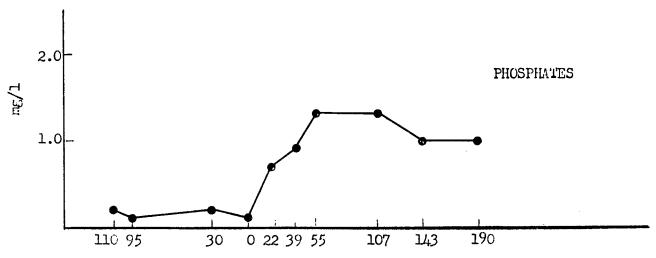


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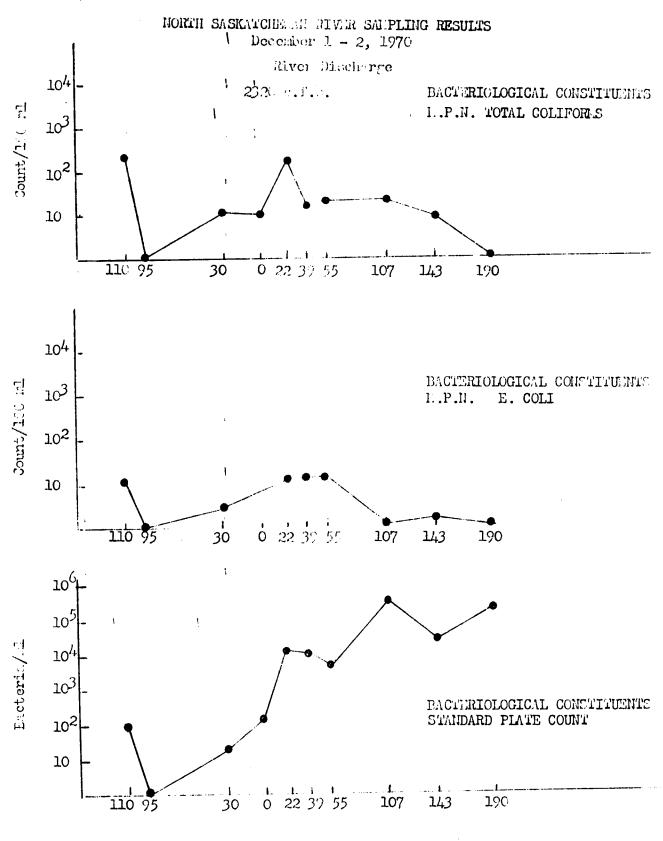






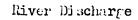


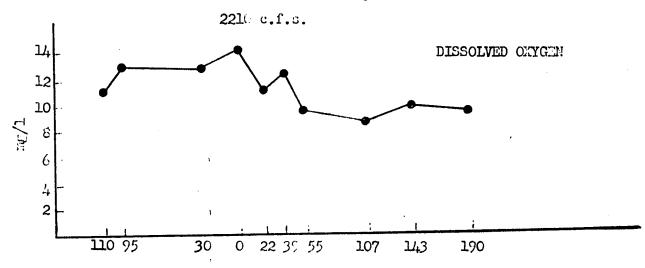
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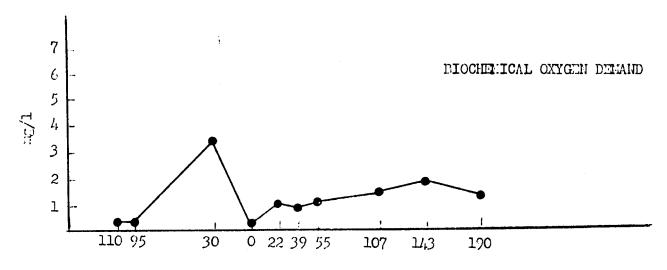


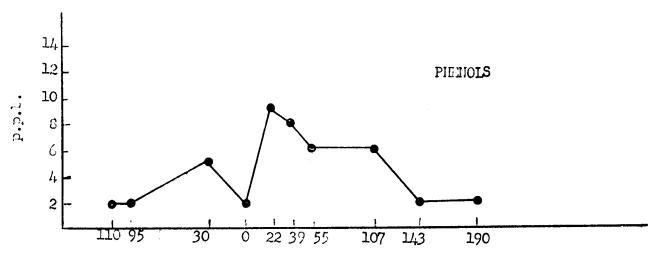
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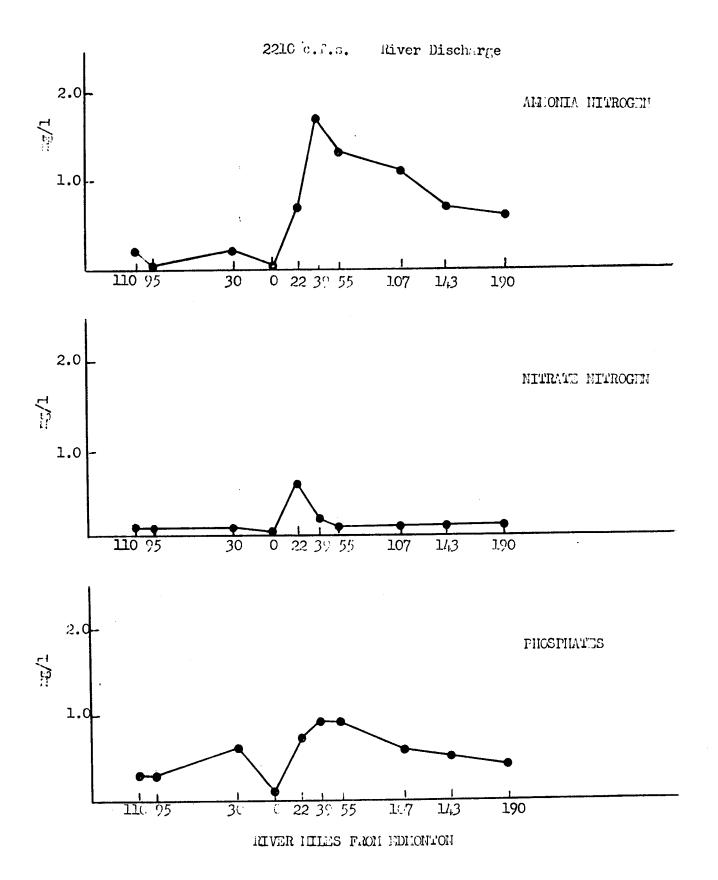


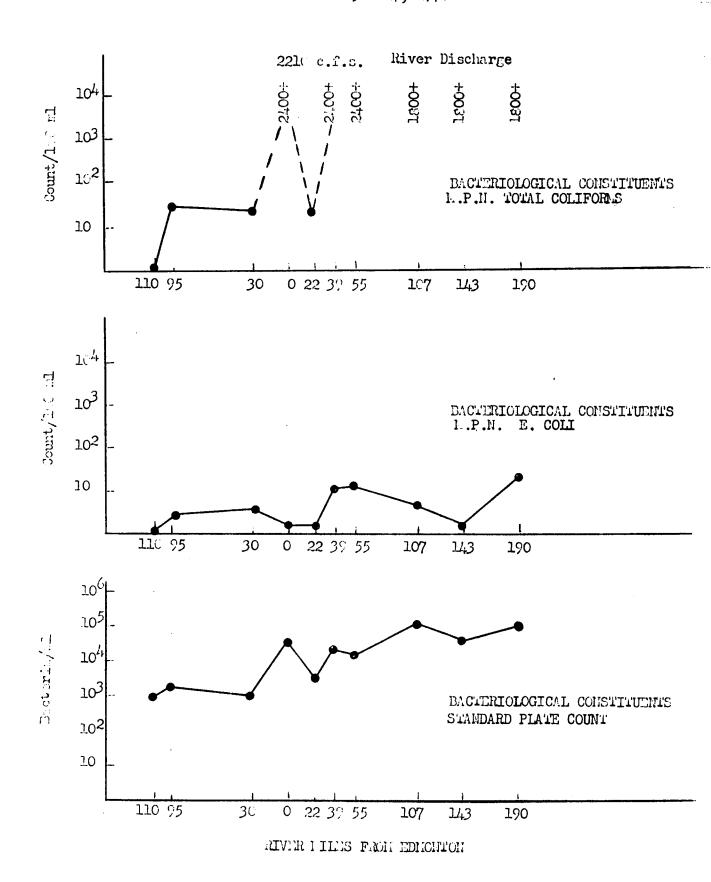




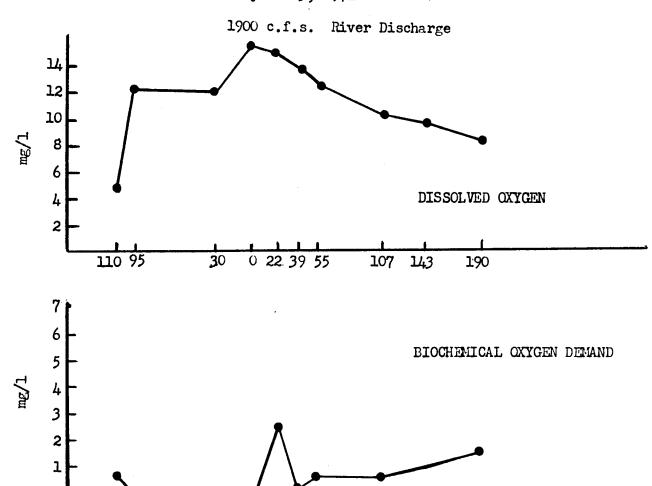


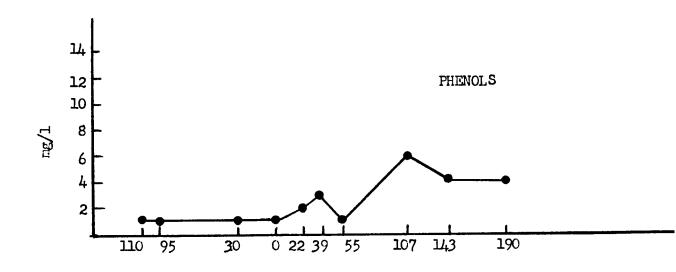
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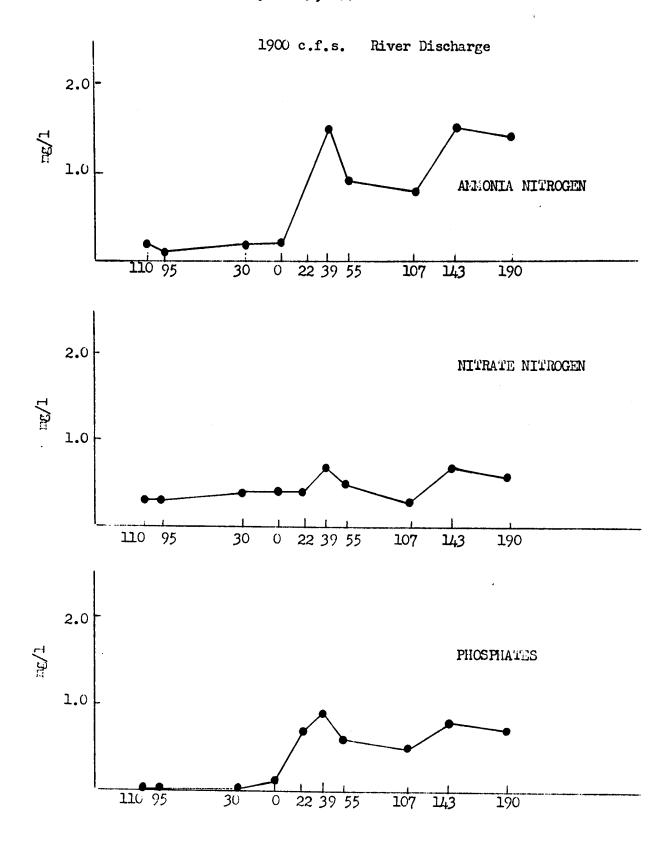




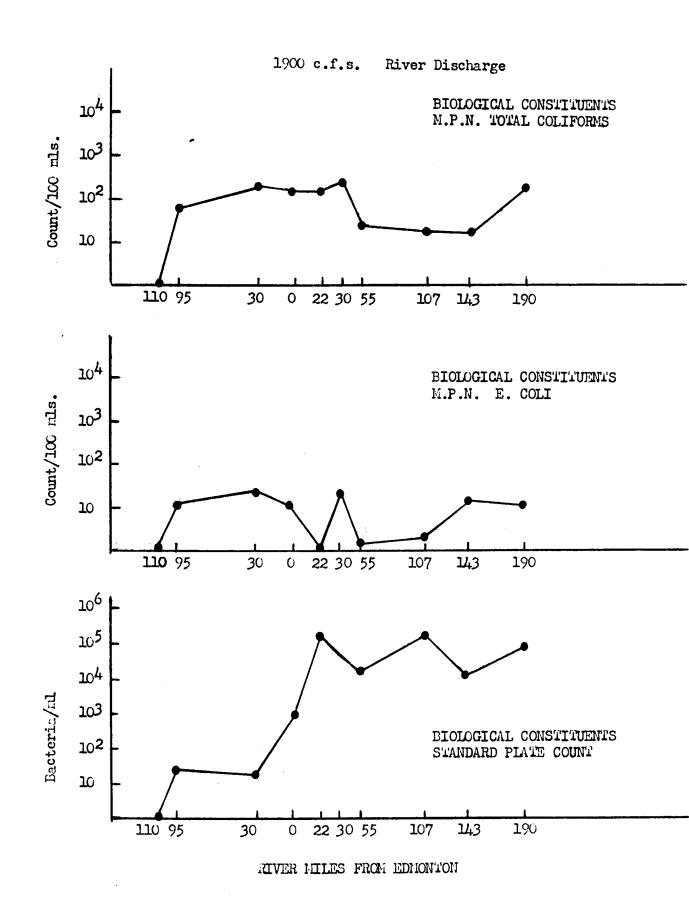
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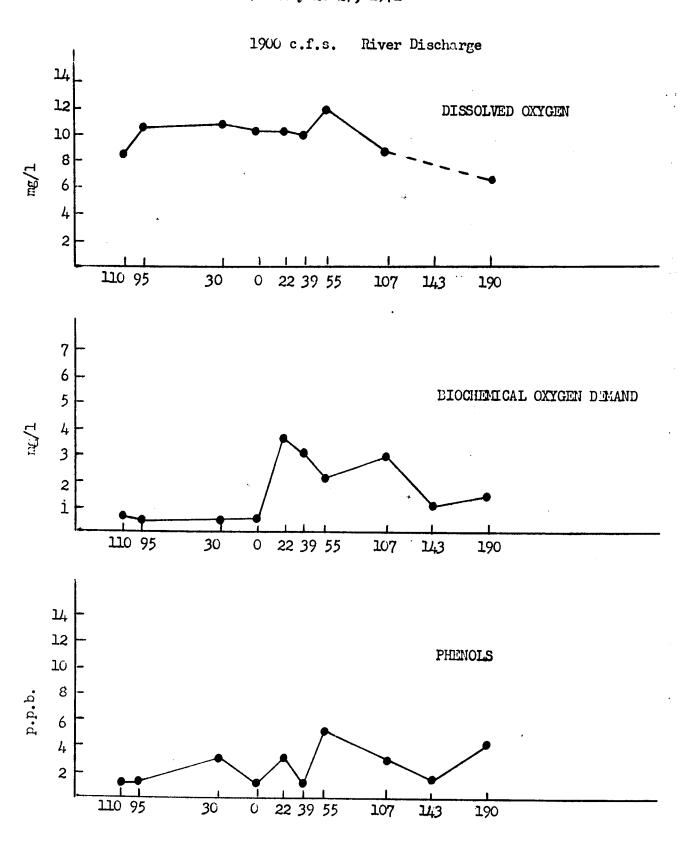
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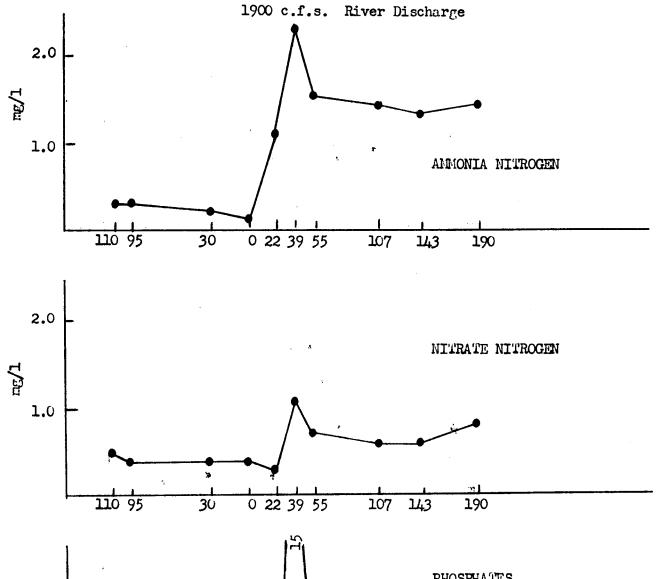
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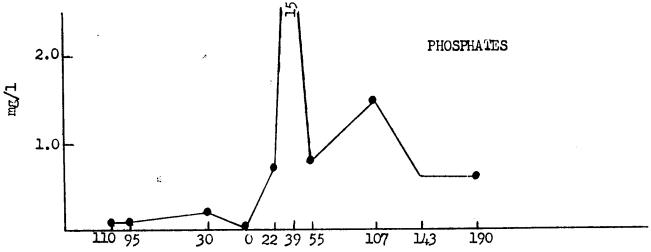


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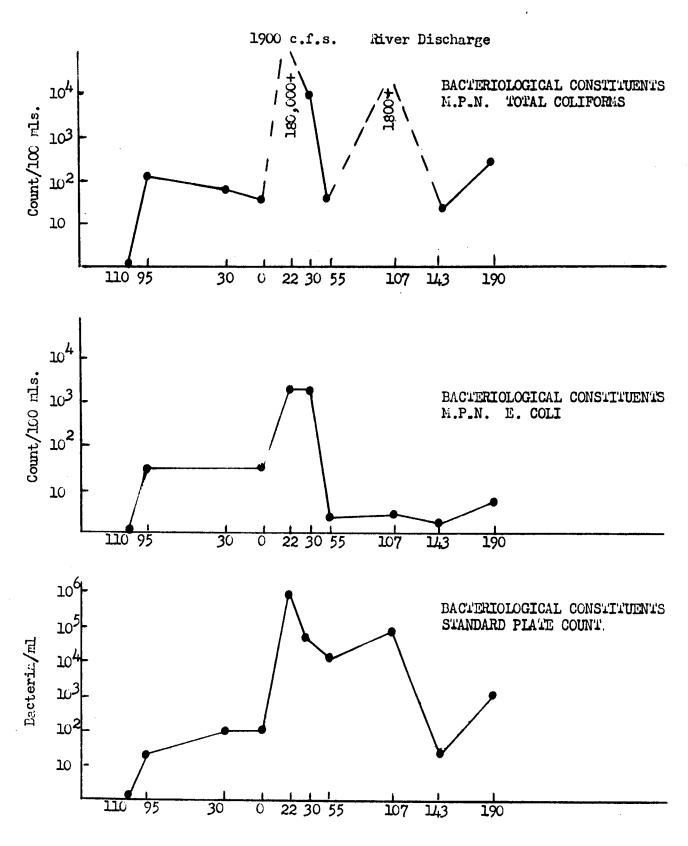
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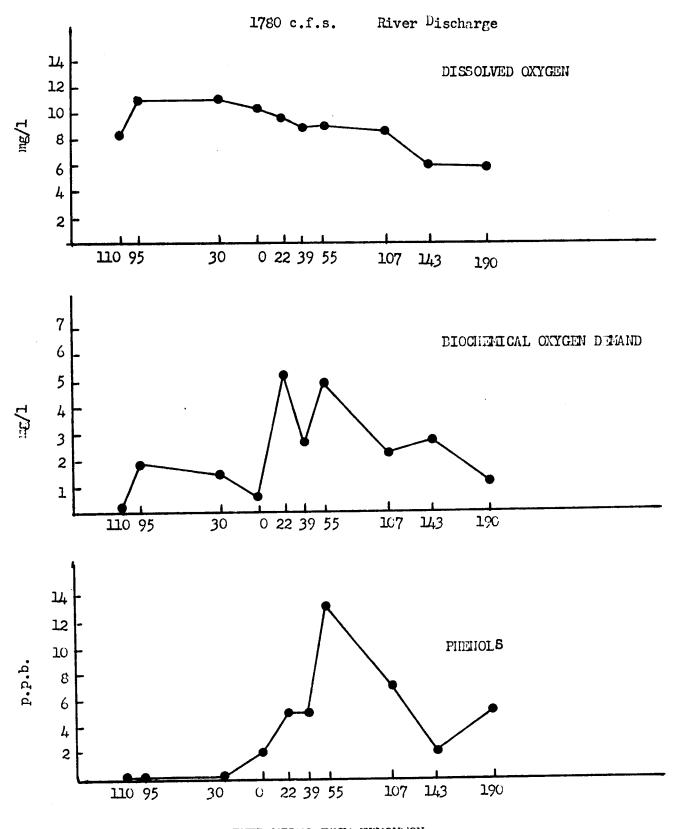
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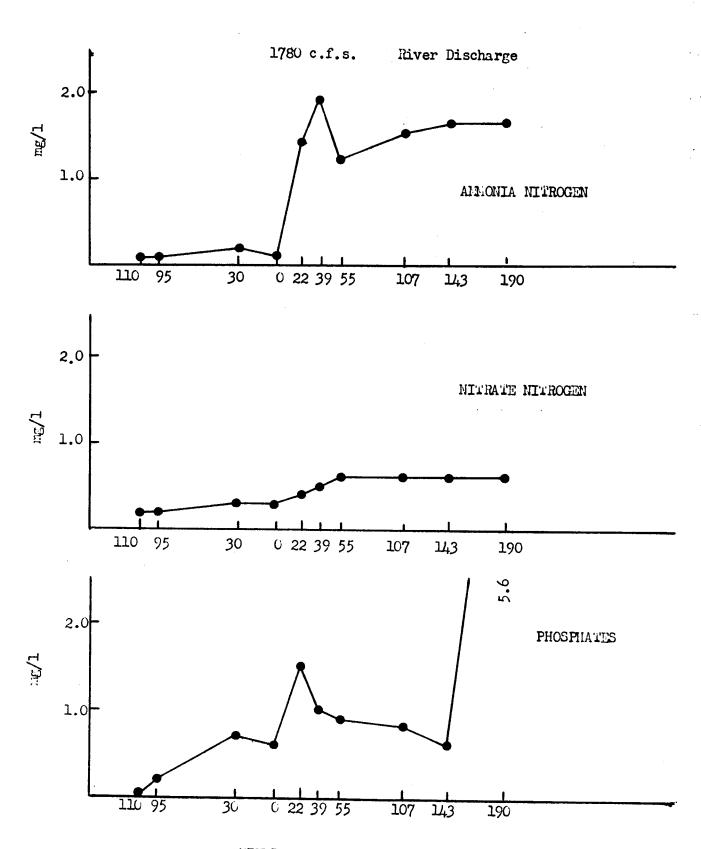
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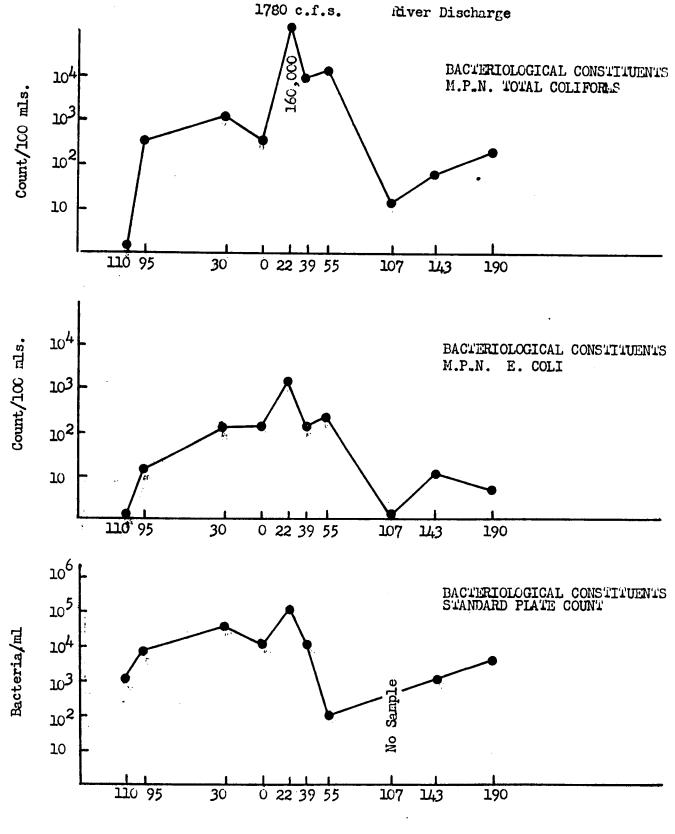
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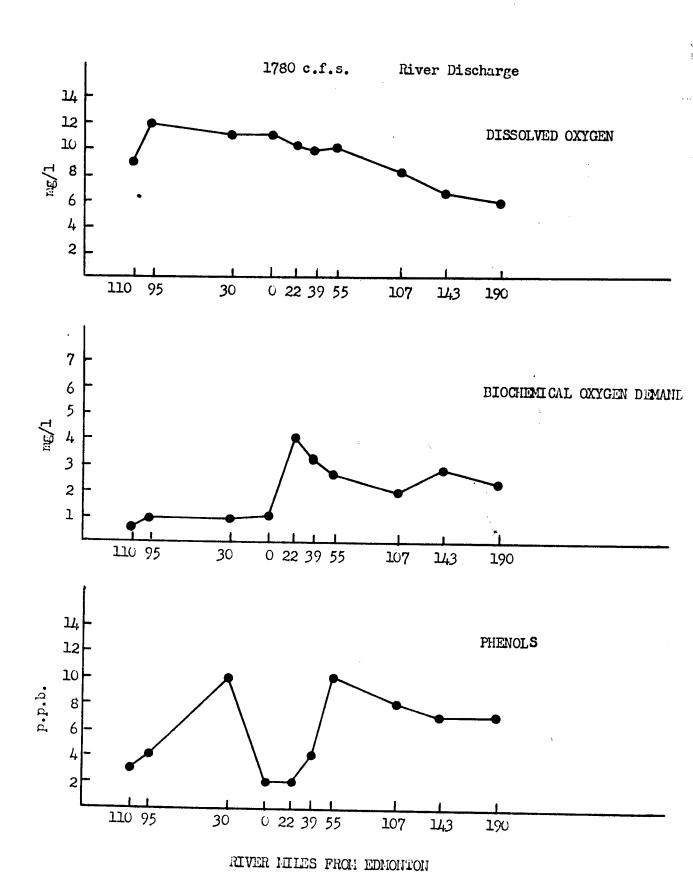
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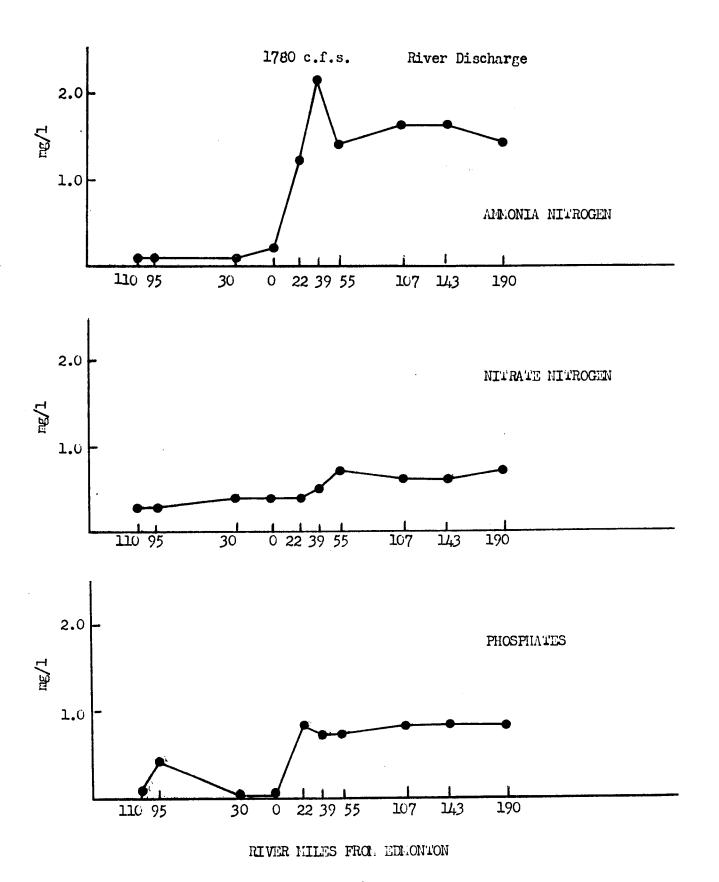


RIVER MILES FROM EDMONTON

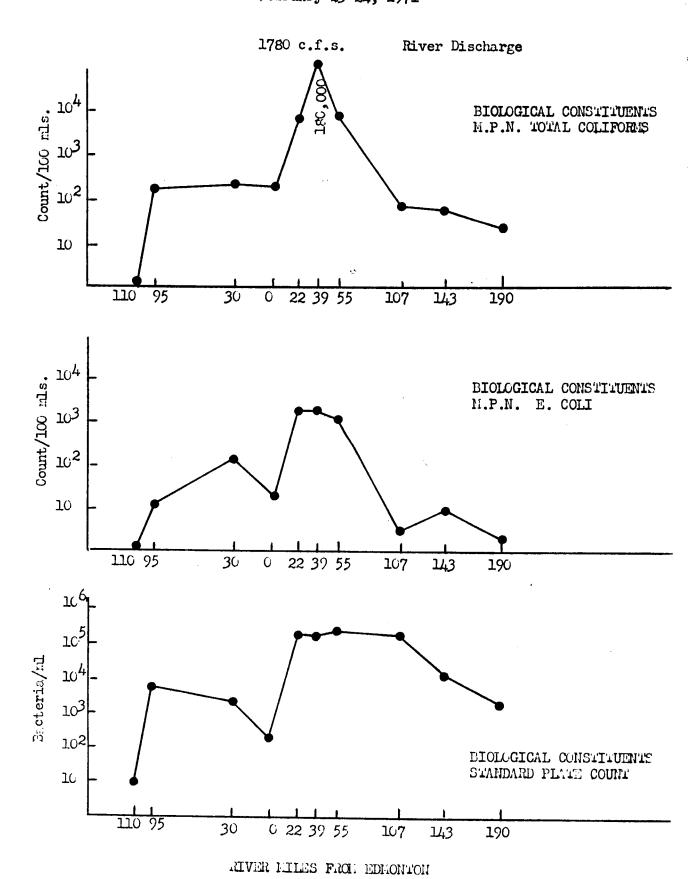
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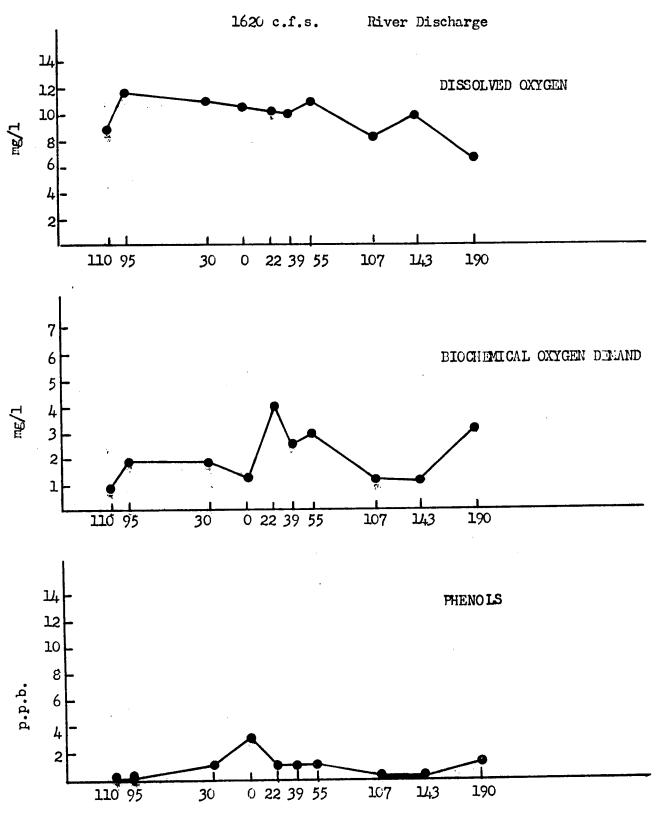
# NORTH SASKATCHEMAN RIVER SAMPLING RESULTS February 23-24, 1971



### NORTH SASKATCHEIAN RIVER SAMPLING RESULTS February 23-24, 1971

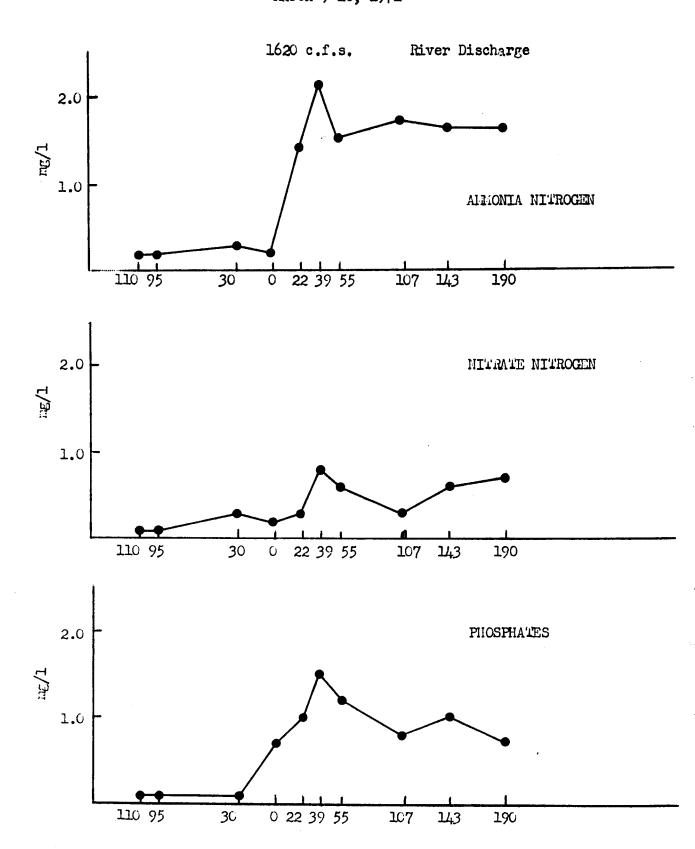


## NORTH SASKATCHEIAN RIVER SAMPLING RESULTS Narch 9-10, 1971



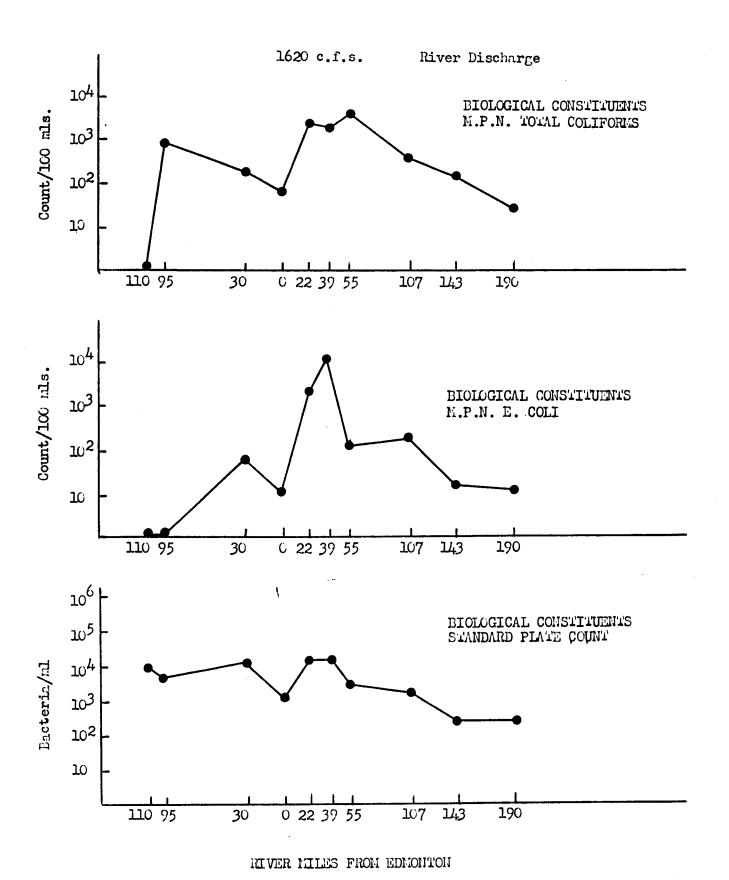
RIVER MILES FROM EDMONTON

### NORTH SASKATCHEAN RIVER SAMPLING RESULTS March 9-10, 1971



RIVER MILES FROM EDMONTON

### NORTH SASKATCHEMAN RIVER SAMPLING RESULTS March 9-10, 1971



### APPENDIX B

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### THERE ARE 365 PIECES OF DATA

THE MINIMUM FLOW IS 1270 CLEIC FEET FER SECOND

1450 CUBIC FEET PER SECCND 14 CF THE TIME THE FLOW IS LESS THAN 5% OF THE TIME THE FLOW IS LESS THAN 1570 CUBIC FEET PER SECOND 1730 CUBIC FEET PEP SECOND 10% OF THE TIME THE FLOW IS LESS THAN LESC CLEIC FEET PER SECCNO 20% OF THE TIME THE FLOW IS LESS THAN 2050 CURIC FEET PER SECURD 3CR OF THE TIME THE FLEW IS LESS THAN 2600 CUBIC FEET PER SECEND 4CR OF THE TIME THE FLOW IS LESS THAN 3140 CUBIC FEET PER SECCENT 50% OF THE TIME THE FLOW IS LESS THAN 4590 CUBIC FEET PER SECOND ECR OF THE TIME THE FLOW IS LESS THAN 736C CLEIC FEET PER SECCND 7CR OF THE TIME THE FLOW IS LESS THAN 8910 CURIC FEET PER SECOND PCR OF THE TIME THE FLOW IS LESS THAN 15200 CLBIC FEET PER SECOND 90% OF THE TIME THE FLOW IS LESS THAN

THE NAXIMUM FLOW IS 53700 CUPIC FEET PER SECOND

# THERE ARE 31C4 PIECES OF DATA THE MINIMUM FLOW IS 1070 CUBIC FEET PER SECOND

18 OF THE TIME THE FLOW IS LESS THAN 1480 CUBIC FEET PER SECOND 5% OF THE TIME THE FLOW IS LESS THAN 1890 CUBIC FEET PER SECOND 1 10% OF THE TIME THE FLOW IS LESS THAN 2170 CUBIC FEET PER SECOND 20% CF THE TIME THE FLOW IS LESS THAN 2460 CUBIC FEET PER SECOND 30% OF THE TIME THE FLOW IS LESS THAN 2790 CUBIC FEET PER SECOND 40% OF THE TIME THE FLOW IS LESS THAN 3340 CUPIC FEET PER SECOND 50% OF THE TIME THE FLOW IS LESS THAN 4160 CUBIC FEET PER SECOND 50% OF THE TIME THE FLOW IS LESS THAN 6150 CUBIC FEET PER SECOND 70% OF THE TIME THE FLOW IS LESS THAN 8600 CUBIC FEET PER SECOND BC% OF THE TIME THE FLOW IS LESS THAN 12200 CUBIC FEET PER SECOND 90% OF THE TIME THE FLOW IS LESS THAN 17800 CUBIC FEET PER SECOND

THE MAXIMUM FLOW IS 91600 CUBIC FEET PER SECOND

The state of the s							
DAY MONTH YEAR	22 APR 1970	4 MAY 1970	22	3 NOV 1970	17 NOV 1970		15 DEC 1970
COMPOSITE OR GRAB SAMPLE	G	G	G	G	G	G	5
INITIAL SAMPLING TIME	1130	1200	1000	1115	1130	1100	1030
TEMPERATURE, DEG.CENT.	5.0	*0.0	10.0	2.5	2.0	0.0	1.0
BARCMETRIC PRES. IN.HG	26.90	*0.00	26.90	27.50	27.10	*0.00	<b>*0.</b> 00
DISSOLVED OXYGEN, MG/L	11.5	*0.0	11.2	13.7	10.2	10.1	11.2
PERCENT SATURATION	100.	*00.	110.	109.	82.	*00.	*00.
BICCHEM. UX. DEMAND MG/L	0.4	1.0	1.1	0.9	0.3	0.4	0.3
HYDROGEN ION CONC., PH	8.2	8.2	8.2	8.3	1.5	7.8	0.3
ALKALINITY MG/L	155	153	120	<b>*</b> 000	*000	*000	*000
THRESHULD ODOR NO., TYPE	2_M	4 M	32 M	4 M	4 M	4 M	2 1
TOTAL RESIDUE MG/L ,	240	270	188	220	226	244	256
IGHITION LOSS MG/L	អូន	138	110	92	50	86	*00
TURBIDITY AS SID2 MG/L	6	4	3	1	ł	4	1
TOTAL HARDNESS MG/L	168	132	116	160	154	1.20	160
CHLORIDES MG/L	1	1	2	1	3	1	1
AMMONIA NITROGEN MG/L	0.1	0.1	0.5	0.1	0.0	0.2	0.2
NITRATE NITROGEN MG/L	0.3	0.1	0.1	0.0	0.1	0.0	0.1
SULFATES AS SO4 MG/L	0	*C0	*00	52	53	54	
TOTAL PHOS.AS PO4 MG/L	0.2	(.1	0.2	0.0	0.0	0.2	0.3
PHENOLS PPB	1	1	2	5	7	6	Ž
DILS & GREASES MG/L	0.5	1.3	1.6	0.9	3.6	2.3	1.3
FLUCRIDES MG/L	0.10	0.15			<b>*.</b> 00	*.00	*.00
COLIFORM M.P.N./100ML.	O	0	4	Ü	C	<b>35</b> 0	Û
MPN OF E COLI/100ML	0	v	0	0	0	14	Ü
STD. PLATE COUNT/ML	200	400	42000	10	10	100	1000
RIVER DISCHARGE C.F.S.	800.	790.	115.	510.	895.	5000.	1030.

<sup>\*</sup> DENOTES DATA NOT AVAILABLE

	DAY MONTH YEAR	12 JAN 1971	26 JAN 1971	9 FEB 1971	23 FEB 1971	9 MAR 1971	
•	COMPOSITE OR GRAB SAMPLE	G	, G	G	G	G	
	INITIAL SAMPLING TIME	1100	1045	1300	1100	1100	
	TEMPERATURE, DEG.CENT.		0.0		0.0	0.5	
	BAROMETRIC PRES. IN.HG	27.20	27.10	26.81	*0.00	27.55	
	- DISSOLVED DXYGEN, MG/L	4.8	8.5	8.4	8.9	8.9	
	PERCENT SATURATION	36.	64.	64.	<b>*</b> 00•	67.	
	BIOCHEM. DX. DEMAND MG/L	1.7	0.6	0.3	0.6	0.8	
	HYDROGEN ION CONC.,PH	7.9	8.0	8.1	7.5	8.3	. *
	ALKALINITY MG/L	149	149	126	141	153	
	THRESHOLD ODOR NO., TYPE	4 M	1 6	4 M	4 M	2 M	
	TOTAL RESIDUE MG/L ,	224	234	256	304	230	
	IGNITION LOSS MG/L	<b>*</b> 00	68	*00	38	110	
	TURBIDITY AS SIDE MG/L	2	1	1	1	1	
	TOTAL HARDNESS MG/L	174	170	170	172		
	CHLORIDES MG/L	4	1	2	U	0	
	AMMONIA NITROGEN MG/L	Ů <b>.</b> 2	0.3	0.1	0.1	0.2	
	NITRATE NITROGEN MG/L	0.3	0.5	0.2	0.3	0.1	
	SULFATES AS SO4 MG/L	42	45	42	40	60	
	TOTAL PHOS.AS PO4 MG/L	0.0	0.1	0.0	0.1	0.1	
	PHENOLS PPB	1	1	0	3	c	
	DILS & GREASES MG/L	2.0	0.7	<b>*</b> •0	0.4	0.9	
	COLIFORM M.P.N./100ML.	c	0	0	0	0	
	MPN OF E CULI/100ML	0		0	0	, 0	
	STD. PLATE COUNT/ML	1	1	1500	10	10000	
	RIVER DISCHARGE C.F.S.	1260.	1000.	805.	800.	990.	

\* DENOTES DATA NOT AVAILABLE

The state of the s							
DAY MONTH YEAR	22 APR 1970	11 JUN 1970	22 SEP 1970	NOV 1970	17 107 1970	DEC 1970	15 DEC 1970
COMPUSITE OR GRAB SAMPLE	(;	G	G	G	G	G	G
INITIAL SAMPLING TIME	1300	1300	1200	1300	1245	1230	1200
TEMPERATURE, DEG.CENT.	3.0	*0.0	10.0	1.5	1.0	0.0	0.0
BAROMETRIC PRES. IN.HG	27.20	*0.00	27.00	27.80	27.40	*0.00	*C.00
DISSOLVED OXYGEN, MG/L	12.4	*() • ()	10.2	13.4	13.3	12.7	12.9
PERCENT SATURATION	101.	<b>*</b> 00.	100.	103.	102.	<b>*</b> 00.	<b>*</b> 00.
BICCHEM. DX. DEMAND MG/L	0.5	0.4	1.4	0.1	0.6	0.7	0.4
HYDROGEN ION CONC.,PH	8.3	8.4	8.3	8.2	3.0	8.0	8.0
ALKALINITY MG/L	141	108	127	<b>*</b> 000	*000	*000	*00C
THRESHOLD ODOR NO., TYPE	4 M	4 CH	4 M	1 M	1 M	2 M	4 M
TOTAL RESIDUE MG/L ,	240	278	220	298	240	248	312
IGNITION LUSS MG/L	33	146	64	120	55	112	<b>≠</b> 1) ∪
TURBIDITY AS SID2 MG/L	26	58	2	3	6	4	3
TOTAL HARDNESS MG/L	164	128	144	200	184	202	200
CHLORIDES MG/L	2	1	0	1	3	1	1
AMMONIA NITROGEN MG/L	0.1	0.2	0.4	0.1	0.2	0.2	0.0
NITRATE NITROGEN MG/L	0.1	0.0	0.0	0.1	0.1	0.0	0.1
SULFATES AS SU4 MG/L	*00	*00	*00	62_	69	<u>6</u> 9	79
TOTAL PHOS.AS PO4 MG/L	0.2	0.3	0.1	0.0	0.2	0.1	0.3
PHENOLS PPB	1	10	2	5	8	5	2
DILS & GREASES MG/L	0.4	0.3	1.0	0.7	1.6	1.7	1.0
FLUORIDES MG/L	0.09	0.15	0.18	*• 60	<b>*</b> •00	<b>*•00</b>	<b>*.</b> 00
COLIFURM M.P.N./100ML.	13	<b>*0000</b> 0	4	4	79	0	47
MPN OF E COLI/100ML	2	*00000	2	Ö	13	0	4
STO. PLATE COUNT/ML	100	*00000	200	1000	100	0	2590

<sup>\*</sup> DENOTES DATA DET AVAILABLE

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DAY MONTH YEAR	12 JAN 1971	KAL	9 FEB 1971	23 FEB 1971	
COMPOSITE OR GRAB SAMPLE	G	G	G	. <b>G</b>	e
INITIAL SAMPLING TIME	1300	1215	1030	1300	1300
TEMPERATURE, DEG.CENT.		0.0	0.0	0.0	0.0
BAROMETRIC PRES. IN.HG	27.40	27.40	27.20	*0.00	27.49
DISSOLVED OXYGEN, MG/L	12.4	10.6	10.9	11.6	11.6
PERCENT SATURATION	93.	79.	82.	*00.	86.
BINCHEM. DX. DEMAND MG/L		0.4	1.8	0.9	1.9
HYDRUGEN ION CONC., PH	8.1	8.3	8.3	7.8	8.1
ALKALINITY MG/L	148	190	141	158	172
THRESHOLD ODOR NO., TYPE	1 M	1 M	1 M	1 M	1 M
TOTAL RESIDUE MG/L ,	227	304	320	384	272
IGNITION LOSS MG/L	*00	<b>7</b> 6	*00	54	138
TURBIDITY AS SID2 MG/L	. 8	1	3	4	8
TOTAL HARDNESS MG/L	184	202	200	206	212
CHLORIDES MG/L	4	1	2	O	1
AMMONIA NITROGEN MG/L	0.1	0.3	0.1	0.1	0.2
NITRATE NITROGEN MG/L	0.3	0.4	0.2	0.3	0.1
SULFATES AS SO4 MG/L		71	73	74	6.2
TOTAL PHOS.AS PO4 MG/L	0.0	0.1	0.2		
PHENOLS PPB	1	1	0	4	U
OILS & GREASES MG/L	1.3	1.0	<b>*.</b> 0	2.7	0.9
COLIFORM M.P.N./100ML.	79	110	540	240	920
MPN OF E COLIZIOOME	11	49	17	17	0
STO. PLATE COUNT/ME	30	30	8000	7500	7000

<sup>\*</sup> DENOTES DATA HOT AVAILABLE

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DAY MONTH YEAR	22 APR 1970	22 SEP 1970	3 NOV 1970	17 NCV 1970	DEC 1971	15 DEC 19 <b>7</b> 0	12 JAM 1971
CUMPUSITE OR GRAB SAMPLE	G	Ç.	G	G	G	G	ß
INITIAL SAMPLING TIME	1430	1530	1430	1430	1530	1630	1500
TEMPERATURE, DEG.CENT.	4.0	12.0	1.5	1.0	0.0	0.0	0.0
BAROMETRIC PRES. IN.HG	27.50	27.10	28.20	27.80	*0.00	*0.00	27.50
DISSULVED GXYGEN, MG/L	12.6	10.0	13.9	13.5	13.0	12.8	12.1
PERCENT SATURATION	105.	102.	105.	102.	<b>*</b> 90.	<b>*</b> 00•	9,0 <b>.</b>
BICCHEM. DX. DEMAND MG/L	2.3	0.9	0.1	0.9	1.1	3.4	0.5
HYDROGEN ION CONC.,PH	8.1	8.5	8.3	8.2	8 • 2	3.1	2.1
ALKALINITY MG/L	126	134	*C00	<b>*</b> 000	*000	<b>*</b> 000	154
THRESHOLD ODOK NO., TYPE	8 %	2 M	1 M	2 M	2 M	1 4	1 "
TOTAL RESIDUE MG/L ,	298	<b>27</b> 0	360	246	30 <b>2</b>	550	240
IGNITION LOSS MG/L	€2	62	202	62	103	90	<b>≯</b> () ु
TUPBIDITY AS SIDE MG/L	30	3	3	8	. 4	57	7
TOTAL HAMDNESS MG/L	160	132	220	200	192	200	201
CHLORIDES MG/L	4	υ	1	2	1	1	1
AMMONIA NITROCEN MG/L	0.2	0.6	0.1	0.1	0.2	0.2	0.2
NITRATE NITROGEN MG/L	0.1	0.0	0.0	0.1	0.0	0.1	0.4
SULFATES AS SU4 MG/L	*00	<b>*</b> 00	<b>7</b> 0	71	62	79	1.0
TOTAL PHOS.AS PO4 MG/L	0.5	0.0	0.1	0.0	0.2	0.6	0.0
PHENOLS PPB	5	0	4	3	9	5	1
OILS & GREASES MG/L	0.9	2.5	0.8	0.9	1.5	1.0	د ∙ ن
FLUCRIDES MG/L	0.09	0.18	*.00	<b>*.</b> 00	<b>*•</b> ()()	*•00	<b>⇒ •</b> ∪ (;
COLIFORM M.P.N./100MI.	4.3	. 0	0	11	13	33	240
MPN OF E COLIZIOOML	22	9	0	i)	4	6	34
STD. PLATE COUNT/ML	650	130	80	100	30	1000	10

	and the second s			· · · · · · · · · · · · · · · · · · ·		
	DAY MONTH YEAR	26 JAN 1971	9 FEB 1971	23 FEB 1971	9 MAR 1971	
	COMPUSITE OR GRAB SAMPLE	G	G	G	G	
	INITIAL SAMPLING TIME	1400	1600	1545	1530	
	TEMPERATURE, DEG.CENT.	0.0	0.0	0.0	0.0	
	BARDMETRIC PRES. IN.HG	27.90	27.30	*0.00	27.45	
	DISSOLVED UXYGEN, MG/L	10.8	10.8	11.0	11.0	
	PERCENT SATURATION	79.	81.	*00.	82.	
•	BIOCHEM. DX. DEMAND MG/L	0.5	1.5	0.9	1.9	
	HYDROGEN ION CONC., PH	8.0	8.2	8.0	8.0	
	ALKALINITY MG/L	179	150	156	180	
	THRESHOLD ODOR NO., TYPE	1 M	4 CH	2 M	2 M	_
	TOTAL RESIDUE MG/L ,	296	328	304	302	
	IGNÍTION LOSS MG/L	72	*00	28	146	
	TURBIDITY AS SIG2 MG/L	6	4	4	2	
	TUTAL HARDNESS MG/L	214	206	206	220	
	CHLORIDES MG/L	1	0	0	2	
	AMMONIA NITROGEN MG/L	0.2	0.2	0.1	0.3	
	NITRATE NITROGEN MG/L	0.4	0.3	0.4	0.3	
	SULFATES AS SU4 MG/L	83	64	70	55	
	TOTAL PHOS.AS PO4 MG/L	0.2	U.7	0.0	0.1	
	PHENOLS PPB	3	0	10	1	
	OILS & GREASES MG/L	0.4	1.5	1.4	0.6	
	CULIFORM M.P.N./100ML.	79	1600	350	240	
	MPN OF E COLIZIOOME	49	170	170	79	
	STD. PLATE COUNT/ML	100	55000	3000	12000	

<sup>\*</sup> DENOTES DATA NOT AVAILABLE

	A CONTRACTOR OF THE CONTRACTOR							
	DAY MONTH YEAR	22 APR 1970	MAY 1970	10 JUN 1970	17 JUN 1970	22 SEP 1970	4 NDV 1970	13 NOV 1970
	COMPOSITE UR GRAB SAMPLE	G	G	G	G	G	G	G
	INITIAL SAMPLING TIME	930	1430	1100	8 3 0	1430	1430	1700
	TEMPERATURE, DEG.CENT.	*0.0	*O.O	*0.0	13.0	13.0	1.0	1.0
	BAROMETRIC PRES. IN.HG	*0.00	*0.00	*0.00	27.80	27.30	27.90	27.40
	DISSOLVED UXYGEN, MG/L	11.5	*0.0	*() <sub>*</sub> ()	<b>*0.0</b>	10.1	13.0	13.1
	PERCENT SATURATION	*00.	<b>*00</b> •	*00.	*00.	104.	98.	101.
	BIOCHEM. DX. DEMAND MG/L	1.7	1.9	1.1	0.9	1.2	0.1	0.7
	HYDRUGEN ION CONC.,PH	8.1	8.5	8.3	9.4	8.6	8.5	3.1
	ALKALINITY MG/L	121	157	122	224	135	170	163
	THRESHOLD ODOR NO., TYPE	100 M	8 M	4 M	32 M	2 M	2 M	2 M
	TOTAL RESIDUE MG/L ,	402	292	400	1359	222	282	246
	IGNITION LOSS MG/L	92	126	174	206	62	56	128
	TURBIDITY AS SIDE MG/L	46	17	84	<b>19</b> 00	4	3	3
	TOTAL HARDNESS MG/L	124	184	220	164	138	200	202
	CHLORIDES MG/L	4	2	3	3	0	4	3
	AMMONIA NITROGEN MG/L	0.3	0.2	0.0	0.3	0.3	0.3	0.6
	NITRATE NITROGEN MG/L	0.5	0.1	0.3	1.2	0.2	0.0	0.2
	SULFATES AS SO4 MG/L	60	78	46	44	55	80	96
	TOTAL PHOS.AS PO4 MG/L	0.7	0.2	0.4	2.9	0.1	0.2	0.1
	PHENOLS PPB	5	2	1	12	1	3	15
	DILS & GREASES MG/L	0.3	2.1	0.4	9.6	3.4	0.9	1.5
	FLUORIDES MG/L	0.10	0.15	0.13	0.16	0.18	0.21	0.01
·	COLIFORM M.P.N./100ML.		*00000	220	36	33	180	41
	MPN OF E COLI/100ML	17	*00000	26	7	17	26	6
•	STD. PLATE COUNT/ML	11000	*00000	1100	2800	50	600	400
	RIVER DISCHARGE C.F.S.				33400.	2670.	1860.	1870.
		and the same of the same of						

<sup>\*</sup> DENOTES DATA NOT AVAILABLE

DAY MONTH YEAR	DEČ 1970	17 DEC 1970	13 JAN 1971	27 JAN 1971	10 FEB 1971	24 FEB 1971	16 MAR 1971
COMPOSITE OR GRAB SAMPLE	Ģ	С	G	C-	G	G	. g
INITIAL SAMPLING TIME	1345	230	1500	1800	1500	1100	1630
TEMPERATURE, DEG.CEMT.	0.0	0.0	0.0	0.0	0.0	0.0	O • 👫
BAROMETRIC PRES. IN.HG	27.70	27.70	27.80	27.30	*0.00	27.00	*0.00
DISSULVED OXYGEN, MG/L	13.1	14.1	15.5	10.2	10.2	10.9	10.00
PERCENT SATURATION	97.	104.	114.	75.	*00.	83.	
BIOCHEM. UX. DEMAND MG/L	0.9	0.3	0.4	0.5	0.6	1.0	1.
HYDROGEN ION COMC., PH	8.1	8.2	7.9	7.8	8.1	8.2	7.
ALKALINITY MG/L	175	174	158	173	112	165	181
THRESHOLD ODOR NO., TYPE	2 14	8 M	8 M	1 11	1 M	1 M	1
TOTAL RESIDUE MG/L ,	280	312	232	288	300	280	354
IGNITION LOSS MG/L	118	32	38	56	74	34	134
TURBIDITY AS SID2 MG/L	2	6	4	2.	4	4	ί.
TOTAL HARDNESS MG/L	206	210	204	203	206	206	21.
CHLORIDES MG/L	1	1	1	i	0	2	¥.
AMMONIA NITROGEN MG/L	0.1	0.0	0.2	0.1	J.1	0.2	0.2
NITRATE NITROGEN MG/L	0.0	0.0	0.4	0.4	0.3	0.4	0
SULFATES AS SU4 MG/L	+6	73	57	73	78	70	66
TOTAL PHOS.AS PO4 MG/L	0.1	0.1	0.1	0.0	0.6	0.0	0.
PHENOLS PPB	7	2	1	1	2	2	3
DILS & GREASES MG/L	1.3	1, . 5	0.7	1.1	0.7	2.0	1.
FLUORIDES MG/L	0.27	0.30	0.30	0.16	*.00	0.44	0.54
COLIFORM M.P.N./100ML.	10	2400+	140	49	540	240	7
MPN OF E CULI/100ML	U	1	11	49	170	27	1
STD. PLATE COUNTYML	120	52000	2000	160	10000	250	1200
RIVER DISCHARGE C.F.S.	2070.	2360.	1850.	1890.	1740.	1470.	<b>*00</b> 00,
CHENICAL OXYGEN DEMAND MG/L			4	6	27	3	5

<sup>\*</sup> DENOTES DATA NOT AVAILABLE

DAY MONTH YEAR	APR 1970	SEP 1970	NUV 1970	18 NGV 1970	DEC 1970	17 DEC 1970	13 JAN 1971
COMPOSITE OR GRAB SAMPLE	G	G	G	G	G	G	G
INITIAL SAMPLING TIME	1230	1300	1330	1615	1300	1600	1300
TEMPERATURE, DEG.CENT.	*0.0	13.0	1.0	1.0	0.0	0.0	0.0
BARDMETRIC PRES. IN.HG	*0.00	27.38	28.00	27.50	27.75	27.90	27.80
DISSOLVED OXYGEN, MG/L	11.9	10.7	14.3	- 13.0	11.5	11.1	14.7
PERCENT SATURATION	<b>*</b> 00.	110.	108.	100.	35.	82.	108.
BIOCHEM. UX. DEMAND MG/L	5.9	3.1	1.6	2.5	1.7	1.3	3.5
HYDROGEN ION CONC.,PH	7.8	8.6	8.0	8.1	7.8	8.1	7.3
ALKALINITY MG/L	121	136	<b>*</b> 0 <b>0</b> 0	<b>*</b> 000	*000	*000	168
THRESHOLD COOR NO., TYPE	50 M	16 M	4 M	4 M	16 M	4 M	, 4 <sub>,</sub> 0
TOTAL RESIDUE MG/L ,	286	378	302	242	274	328	247
IGNITION LOSS MG/L	98	166	160	134	92	*00	<b>*</b> 00
TURBIDITY AS SHOZ MG/L	34	6	4	3	1	3	10
CHLORIDES MG/L	4	4	7	8	2	3.	1
AMMONIA NITROGEN MG/L	1.0	0.7	1.1	1.2	0.9	0.7	6.3
NITRATE NITROGEN MG/L	0.2	0.0	0.1	0.1	0.0	0.6	0.4
SULFATES AS SO4 MG/L	56	48	<b>*0</b> 0	*00	<b>*</b> 00	*00	*00
TOTAL PHOS.AS PO4 MG/L	0.7	0.8	0.6	0 • 8	0.7	0.7	0.7
PHENOL S PPB	2	5	3	14	4	9	2
OILS & GREASES MG/L	0.6	1.9	0.9	2.5	4.6	1.2	1.0
FLUORIDES MG/L	0.13	0.20	<b>*•0</b> 0	<b>*.</b> 00	<b>*.</b> 00	<b>*.</b> 00	*.00
COLIFORM M.P.N./100ML.	18000	1800	350	23	250	31	140
MPN UF E COLI/100ML	90	38	6	6	16	1	0
STD. PLATE COUNT/ML	4800	150000	4300	200000	18000	5000	20000c
DETERGENTS MG/L CHEMICAL OXYGEN DEMAND MG/L	0.00	0.30	0.30	*0.00	0.20	*0.00	0.50 <b>7</b> .

DAY MONTH YEAR	27 JAN 1971	FĒB	24 FEB	10 MAR 1971	140.46	
COMPOSITE OR GRAB SAMPLE	E G	G	G	G		
INITIAL SAMPLING TIME	1600	1400	1430	1500		
TEMPERATURE, DEG.CENT.	0.0	0.0	0.0	0.0		
BAROMETRIC PRES. IN.HG	27.90	*0.00	2 <b>7.0</b> 0	*0.00		
DISSOLVED OXYGEN, MG/L	10.2	9.5	10.0	10.1		
PERCENT SATURATION	75.	*0C•	76.	<b>*00</b> .		
BIOCHEM. OX. DEMAND MG/L	. 3.6	5.2		4.0		•••
HYDRUGEN ION CONC.,PH	7.9	8.0	8.1	7.7		
ALKALINITY MG/L	176	280		181		
THRESHOLD ODUR NO., TYPE	4 C	4 C		1 C		
TOTAL RESIDUE MG/L .	324			336		•
IGNITION LOSS MG/L	62	<b>*00</b>	78	94		
TURBIDITY AS SID2 MG/L	16	12		7		
CHLORIDES MG/L	5		_	<u> </u>		
AMMONIA NITROGEN MG/L	1.1			1.4	• •	
NITRATE NITRUGEN MG/L	C.3			0.3		}
SULFATES AS SO4 MG/L			•	70		
TOTAL PHUS.AS PO4 MG/L	0.7			1.0		
PHENOLS PPB	3	5		1	** **	
OILS & GREASES MG/L	0.8		-	0.3		
COLIFORM M.P.N./100ML.				3400		
MPN OF E COLI/100ML	2600					
STD. PLATE COUNT/ML				2700		
DETERGENTS MG/L	*0.00			18000		
CHEMICAL OXYGEN DEMAND MG/L				0.20		
	14.	40.	10.	7.		

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	DAY MONTH YEAR	17 JUN 1970	23 JUN 1970	15 JUL 1970	22 APR 1970	22 SEP 1770	4 NOV 1970	18 NOV 1970
	COMPOSITE OR GRAB SAMPLE	G	G	G	G	G	G	G
	INITIAL SAMPLING TIME	1230	1100	*000	1100	1030	1230	1530
	TEMPERATURE, DEG.CENT.	14.0	*0.0	*0.0	*0.0	11.0	1.0	1.0
	BAROMETRIC PRES. IN.HG	28.00	*0.00	*0.00	*0.00	27.45	28.10	21.50
	DISSOLVED DXYGEN, MG/L	*0.0	*0.0	*0.0	10.3	9.4	13.1	13.0
	PERCENT SATURATION	*00.	*00.	*00.	<b>*</b> 00.	32.	98.	100.
	BIOCHEM. UX. DEMAND MG/L	4.7	4.0	1.0	3.5	4.0	1.6	2.7
	HYDROGEN TON CONC.,PH	9.2	8.3	8.6	7.8	8.4	8.1	8.1
	ALKALINITY MG/L	221	132	130	105	137	*000	*000
	THRESHOLD ODUR NO., TYPE	16 M	8 _ <b>C</b>	32 C	50 M	8 M	8 M	8 M
	TOTAL RESIDUE MG/L ,	1242	448	332	300	288	314	318
	IGNITION LOSS MG/L	320	126	120	90	118	98	100
	TURBIDITY AS SIU2 MG/L	2400	525	600	44	6	4	3
	CHLORIDES MG/L	9	9	9	6	3		12
	AMMONIA NITROGEN MG/L	0.7	0.6	0.2	1.4	1.2	2.1	1.7
	NITRATE NITRUGEN MG/L	0.6	0.2	0.2	0.7	0.3	1.1	0.2
	SULFATES AS SO4 MG/L	38	34	34	64	54	<b>*</b> 00	<b>*</b> 00
_	TOTAL PHOS.AS PO4 MG/L	0.1	0.3	0.2	1.2	0.6	0.6	0.9
	PHENOL S PPB	16	1	2	6	4	3	12
	DILS & GREASES MG/L	· * • O	6.0	2.7	*•0	<b>*•</b> 0	<b>*</b> • 0	<b>*.</b> 0
	FLUORIDES MG/L	<b>*.</b> 00	*.00	*.00	<b>*.</b> 00	<b>*.</b> 00	<b>*.</b> 00	0.02
	COLIFORM M.P.N./100ML.	45	*00000	35000	18004	920	1800+	19
	MPN OF E COLI/100ML	1	*00000	930	15	9	12	3
	STD. PLATE COUNT/ML	32000	*00000	600000	3800	220000	2600	95000

 			* **				
DAY MONTH YEAR	DEC 1970	17 DEC 1970	13 JAN 1971	27 JAN 1971	10 FEB 1971	24 FEB 1971	1 MA 1971
COMPOSITE OR GRAB SAMPLE	G	G	G	G	G	G	3
 INITIAL SAMPLING TIME	1200	1400	1230	1400	1230	1300	1300
TEMPERATURE, DEG.CENT.	0.0	0.0	0.0	0.0	0.0	0.0	0.
BAROMETRIC PRES. IN.HG	27.75	27.92	27.80	28.01	*0.00	27.10	*C.O.
DISSOLVED OXYGEN, MG/L	10.0	12.2	13.6	9.9	8.7	9.6	10.
 PERCENT SATURATION	74.	90.	100.	72.	<b>*</b> 00.	73.	*00
BIOCHEM. DX. DEMAND MG/L	2.0	0.9	1.2	3.0	2.6	3.2	2.
HYDROGEN ION CONC., PH	7.8	8.0	7.1	7.4	7.8	8.1	7.
ALKALINITY MG/L	<b>*000</b>	*000	166	176	211	168	182
 THRESHOLD ODOR NO., TYPE	M 8	4 14	2 M	4 C	4 CH	2 <sub></sub> C	1
TOTAL RESIDUE MG/L ,	294	294	254	336	386	372	378
IGNITION LOSS MG/L	126	34	80	68	142	154	8
TURBIDITY AS SID2 MG/L	1	*00	7	2	9	7	· ·
 CHLORIDES MG/L	4	2	1	6	e e	8	2
AMMONIA NITROGEN MG/L	1.4	1.7	1.5	2.3	1.9	2.1	2.
NITRATE NITROGEN MG/L	0.1	0.2	0.7	1.1	0.5	0.5	0.1
SULFATES AS SU4 MG/L	#00	*00	*00	<b>*</b> 00	*00	*00	7
TOTAL PHOS.AS PO4 MG/L	0.9	0.9	0.9	15.0	1.0	0.7	1.5
PHENOL S PPB	6	8	3	1	5	4	
OILS & GREASES MG/L	*•0	*.O	1.0	0.8	0.6	1.5	0.3
FLUORIDES MG/L	0.28	0.30	0.30	0.19	0.28	0.42	0.4
COLIFORM M.P.N./100ML.	22	2400	400	9200	9200	180000	1400u
MPN OF E COLI/100ML	17	12	36	2400	170	2800	210^
STD. PLATE COUNT/ML	10000	33000	62000	65000	13000	200000	20000

<sup>\*</sup> DENOTES DATA NOT AVAILABLE

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DAY MONTH YEAR	22 APR 1970	22 SEP 19 <b>7</b> 0	NOV 1970	18. NCV 1970	DEC 1970	17 DEC 1970	13 JAN 1971	
COMPOSITE UR GRAB SAMPLE	G	G	G	G	G	G	G	
INITIAL SAMPLING TIME	930	1000	1130	1200	1100	1230	110e	
TEMPERATURE, DEG.CENT.	*0.0	11.0	0.5	1.0	0.0	0.0	0.0	
BARDMETRIC PRES. IN. HG	*0.00	27.50	28.10	27.40	27.60	27.90	27.30	
DISSOLVED DXYGEN, MG/L	10.3	9.7	12.5	12.7	9.5	9.3	12.7	
PERCENT SATURATION	<b>*00.</b>	95.	92.	98.	70.	68.	94.	
BIOCHEM. DX. DEMAND MG/L	2.7	2.5	2.4	2.0	3.3	1.1	1.0	
HYDROGEN ION CONC., PH	7•8	8.7	0.8	8.2	7.8	0.9	8.1	
-ALKALINITY MG/L	97	136	*000	<b>*</b> 000	<b>*</b> 000	*000	175	
THRESHOLD ODOR NO., TYPE	100 i	M 8	16 M	4 M	16 M	32 C	1 C	
TOTAL RESIDUE MG/L ,	298	254	412	338	280	344	<b>27</b> 3	
IGNITION LOSS MG/L	106	124	108	130	80	<b>*</b> 00	*0(	
TURBIDITY AS SID2 MG/L	18	4	39	3	ı	1	7	
CHLORIDES MG/L	5	6	12	3	5	5	4	
AMMONIA NITRUGEN MG/L	1.4	0.6	1.6	1.1	1.6	1.3	0.5	
NITRATE NITROGEN MG/L	0.4	0.1	0.6	0.2	0.0	0.1	0.5	
SULFATES AS SO4 MG/L	66	53	*00	*00	<b>*</b> 00	*00	<b>*</b> (1)	
TOTAL PHUS.AS PL4 MG/L	1.0	1.0	0.6	0.8	1.3	0.9	0.6	
PHENOLS PPB	9	4	5	13	11	<i>6</i>	1	
OILS & GREASES MG/L	*•()	<b>* •</b> 0	<b>*</b> • O	* • ()	<b>* •</b> 0	<b>*</b> • 0	1.0	
COLIFORM M.P.N./100ML.	81	350	170	· 11	28	2400+	35	
MPN OF E COLI/100ML	13	0	6	U	17	14	1	
STD. PLATE COUNT/ML	2000	16000	25000	1500	6800	23000	23000	

DAY MONTH YEAR	27 JAN 1971	10 FEB 1971	FĒB	10 MAR 1971
COMPOSITE OR GRAD SAMPLE	G	G	G	G
INITIAL SAMPLING TIME	1100	1145	1200	1200
TEMPERATURE, DEG.CENT.	0.0	0.0	0.0	0.0
BAROMETRIC PRES. IN.HG	28.03	*0.00	27.20	*0.00
DISSOLVED OXYGEN, MG/L	11.5	8.8	9.9	10.8
PERCENT SATURATION	84.	*00.	75.	<b>*</b> 00.
BIOCHEM. DX. DEMAND MG/L	2.1	4.9	2.6	2.9
HYDROGEN ION CONC.,PH	7.8	7.8	8.1	7.6
ALKALINITY MG/L	176	196	168	192
THRESHOLD ODOR NO., TYPE	2 M	2 C	1 C	1 M
TOTAL RESIDUE MG/L ,	356	352	312	338
IGNITION LOSS MG/L	62	*00	110	134
TURBIDITY AS SID2 MG/L	2	3	3	2
CHLORIDES MG/L	19	.9	9	20
AMMONIA NITROGEN MG/L	1.5	1.2	1.4	1.5
NITRATE NITROGEN MG/L	0.7	0.6	0.7	0.6
SULFATES AS SO4 MG/L	<b>*0</b> 0	<b>*</b> 00	<b>*</b> 00	85
TOTAL PHOS.AS PO4 MG/L	0.8	0.9	0.7	1.2
PHENOLS PPE	5	13	10	1
OILS & GREASES MG/L	<b>*</b> • ()	0.7	0.4	0.3
COLIFORM M.P.N./100ML.	54	18000	8100	5400
MPN OF E COLIVIOOME	3	320	1500	150
STD. PLATE COUNT/ML	12000	100	300000	4000

<sup>\*</sup> DENOTES DATA MOT AVAILABLE

26 JAN 1971
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<sup>\*</sup> CENGTES DATA NOT AVAILABLE

DAY MCNTH YEAR	10 FEB 1971	24 FEB 1971	10 MAR 1971	•
COMPOSITE OR GRAB SAMPLE	G	G	G.	
INITIAL SAMPLING TIME	93C	1030	1030	•
TEMPERATURE, DEG.CENT.	0.0	C.O	0.0	
BARDMETRIC PREST IN HG	27.70	27.30	27.80	
DISSCLVED OXYGEN, MG/L	7.8	7.5	8.1	
PERCENT SATURATION	58.	56.	60.	
BICCHEM- OX. DEMAND MG/L	2.3	1.9	1.2	
HYCROGEN ICH CONC., PH	7.8	8.0	8.2	
ALKALINITY MG/L	191	180	197	
THRESHOLD GDGR NO., TYPE	2 M	2 C	1 C	
TOTAL RESIDUE MG/L ,	404	316	314	
IGNITION LOSS MG/L	*0C	76	122	
- TURBICITY AS SIC2 MG/L	3	.2	2	
CHLCRIDES MG/L	22	19	16	
AMMONIA NITROGEN MG/L	1.5	1.6	1.7	
NITRATE NITROGEN MG/L	0.4	0.6	0.3	
- SULFATES AS SO4 MG/L	.*00	*C0	75	• •
TOTAL PHOS.AS PC4 MG/L	1.1	C.8	C • 8	1
PHENCLS PPB	7	8	0	
DIES & GREASES MG/L	<b>*</b> •0	2.4	1.0	
COLIFORM M.P.N./100ML.	17	81	540	
MPN OF E COLI/1COML	C		220	
STC. PLATE CCUNT/ML	11 <b>0</b> 0000	190000	2100	

<sup>\*</sup> CENCTES CATA NCT AVAILABLE

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DAY MONTH YEAR	23 APR 1970	15 JUL 1970	23 SEP 1970	4 NOV 1970	19 107 1970	DEC 19 <b>7</b> 0	16 DEC 1970
COMPOSITE OR GRAB SAMPLE	G	G	C	G	G	G	3
INITIAL SAMPLING TIME	1100	1400	1200	1030	1400	1200	1145
TEMPERATURE, DEG.CENT.	<b>*0.</b> 0	*C•0	9.0	1.0	1.0	0.0	0.0
BARDMETRIC PRES. IN.HG	<b>*0.00</b>	*0.C0	27.71	*C.00	27.76	28.00	28.00
DISSOLVED OXYGEN, MG/L	12.1	*O.O	11.9	13.1	<b>*0.0</b>	8.3	9.5
PERCENT SATURATION	*00.	<b>*</b> 00.	111.	*0C•	<b>*00</b> •	61.	70.
BIOCHEM. DX. DEMAND MG/L	4.0	1.2	4.3	3.9	2.2	1.1	1.7
HYDROGEN ION CONC., PH	7.8	8.5	8.7	7.7	8.3	8.1	8.2
ALKALINITY MG/L	110	161	130	*000	*000	*000	<b>*0</b> 00
THRESHOLD ODOR NO., TYPE	100 M	4 C	16 M	M s	B M	4 M	۶ c
TOTAL RESIDUE MG/L .	304	392	290	296	380	348	336
IGNITION LOSS MO/L	100	112	92	86	108	124	*0¢
TURBIDITY AS SID2 MG/L	38	650	17	3	3	1	3
CHLORIDES MG/L	10	5	8	15	17	17	5
AMMONIA NITROGEN MG/L	1.2	0.4	0.0	1.0	1.4	2.4	0.7
NITRATE NITROGEN MG/L	0.5	0.3	0.1	0.6	0.4	0.1	0.1
SULFATES AS SO4 MG/L	60	38	52	<b>*00</b>	<b>*</b> 00	*00	<b>*</b> 00
TOTAL PHOS.AS PC4 MG/L	0.7	() • 4	C • 4	0.5	0.6	1.0	0.5
PHENOLS PPD	11	2	3	4	4	12	2
OILS & GREASES MG/L	8.0	1.0	7.7	8.0	2.4	3.4	1.9
FLUORIDES MG/L	0.15	0.21	().14	<b>*•</b> 00	<b>*•</b> ()()	<b>* •</b> 0()	36.*
COLIFORM M.P.N./100ML.	240	*00000	7	4	13	4	1850 <sup>+</sup>
MPN OF E COLI/100ML	27	*000C0	2	2	7	2	1
STD. PLATE COUNT/ML	11000	*00000	5000	620000	300	45000	63000

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DAY Month Year	13 JAN 1971	26 JAN 1971	10 FEB 1971	F53 1971	10 MAR 1971
COMPOSITE OR GRAB SAMPLE	G	G	G	. G	G
INITIAL SAMPLING TIME	1300	1100	1200	1200	1230
TEMPERATURE, DEG.CENT.	0.0	0.0	0.0	0.0	0.0
BAROMETRIC PRES. IN.HG	28.10	28.60	27.70	27.50	28.00
DISSOLVED OXYGEN, MC/L	9∙ ક	*() • ()	5.7	6.4	9.7
PERCENT SATURATION	71.	*00.	42.	48.	71.
BIOCHEM. OX. DEMAND MG/L	1.8	1.0	2.7	1.8	1.1
HYDROGEN ION CONC.,PH	8.1	7.9	0.8	8.0	მ∙0
ALKALINITY MG/L	187	184	176	173	194
THRESHOLD ODOR NO., TYPE	2 C	2 C	2 M	1 M	1 C
TOTAL RESIDUE MG/L ,	309	366	408	370	360
IGNITION LOSS MG/L	*00	84	*00	100	138
TURBIDITY AS SID2 MG/L	9	7	3	2	2
CHLORIDES MG/L	23	20	2	25	0
AMMONIA NITROGEN MG/L	1.5	1.3	1.6	1.6	1.6
NITRATE NITROGEN MG/L	0.7	0.6	0.6	0.6	0.6
SULFATES AS SO4 MG/L	*00	*0U	<b>*</b> U0	*00	72
TOTAL PHOS.AS PU4 MG/L	9.0	0.6	0.6	0.8	1.0
PHENOLS PPB	4	1	2	7	o
OILS & GREASES   MG/L	0.3	1.5	* • ()	2.0	1.2
COLIFORM M.P.N./100ML.	24	22	79	72	130
MPN OF E COLIVIOOME	15	2	14	10	22
STD. PLATE COUNTYME	12000	2700	1000	20000	400

<sup>\*</sup> DENOTES DATA NOT AVAILABLE

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	DAY MONTH YEAR	23 APR 1970	5 MAY 1970	17 JUN 1970	23 SEP 1970	40V 1970	19 NOV 1970	DEC 19 <b>7</b> 0
	COMPOSITE OR GRAB SAMPLE	G	G	G	G	G	G	G
	INITIAL SAMPLING TIME	1300	1400	1400	<b>*000</b>	1430	*000	1430
	TEMPERATURE, DEG.CENT.	0.0	0.0	15.0	9.0	1.0	1.0	0.0
	BAROMETRIC PRES. IN.HG	*0.00	*0.00	*0.00	27.80	*0.00	27.80	28.10
	DISSOLVED OXYGEN, MG/L	10.5	*0.0	*0.0	10.2	*0.0	*0.0	9.0
	PERCENT SATURATION	*CO.	*00.	<b>*00</b> •	95.	*00.	<b>*</b> 00•	66.
	BIOCHEM. DEMAND MG/L	5.3	2.8	4.2	3.0	3.9	0.7	0.4
	HYDROGEN ION CONC. PH	0.8	8.5	8.9	8.7	3.0	8 • 4	8.0
	ALKALINITY MG/L	111	145	138	127	169	191	203
	THRESHOLD UDOR NO., TYPE	50 M	8 M	8 M	8 M	16 M	8 M	4 M
	TOTAL RESIDUE MG/L ,	362	308	348	296	396	<b>34</b> 8	386
	IGNITION LOSS MG/L	126	82	96	80	134	86	122
	TURBIDITY AS SID2 MG/L	47	18	52	y,	4	3	3
	TUTAL HARDNESS MG/L	122	158	170	126	210	2 30	238
	CHLORIDES MG/L	12	1	3	22	28	33	15
	AMMONIA NITROGEN MG/L	1.1	0.5	0.7	0.0	1.3	1.4	1.9
	NITRATE NITROGEN MG/L	0 • 4	0.3	0.2	0.1	0.6	0.6	0.1
	SULFATES AS SO4 MG/L	58	68	38	46	74	84	89
•	TOTAL PHOS.AS PO4 MG/L	0.8	0.6	0.4	0.3	0.9	0.7	1.0
	PHENOLS PPB	6	2	12	0	4	4	11
	OILS & GREASES MG/L	0.9	1.7	0.3	7.0	1.0	1.5	1.0
	FLUORIDES MG/L	0.17	0.13	0.26	0.18	0.25	0.01	*.00
•	COLIFORM M.P.N./100ML.	27	6	72	7	4	7	0
	MPN OF E CULI/100ML	12	0	0	4	0	0	o
	STD. PLATE COUNT/ML	6000	7000	46000	3500	570000	400	370000
	DETERGENTS MG/L							
	RIVER DISCHARGE C.F.S.						*0000•	

<sup>\*</sup> DENOTES DATA NOT AVAILABLE

DAY MONTH YEAR	16 DEC 1970	13 JAN 1971	26 JAN 1971	10 FEB 1971	24 FEB 1971	10 MAR 1971	
COMPOSITE OR GRAB SAMPLE			G		G	G	
INITIAL SAMPLING TIME			1430	1400	1345	1400	
TEMPERATURE, DEG.CENT.						· · · · · · · · · · · · · · · · · · ·	-
BAROMETRIC PRES. IN.HG	28.00	28.20	28.60	27.70	27.50	28.00	
DISSOLVED OXYGEN, MG/L	9.2	8.5	6.1	5.6	5.7	6.5	
PERCENT SATURATION	67.	62.	44.	41.	42.	48.	
BIOCHEM. OX. DEMAND MG/L	1.3	2.5	1.3	1.2	2.2	3.1	
HYDROGEN ION CONC.,PH	8.2	8.1	7.9	7.7	3.0	7.9	
ALKALINITY HG/L	199	202	179	181	179	188	
THRESHOLD ODOR NO., TYPE	4 M	8 M	2 ε	1 M	1 C	1 C	
TOTAL RESIDUE MG/L ,	306		374	464	420	362	
IGNITION LOSS MG/L	108	54	90	96	93	116	
TURBIDITY AS SID2 MG/L	1	10	3	9	2	2	
TOTAL HARDNESS MG/L	225	226	222	214	212	204	
CHLORIDES MG/L	13	13	26	48	44	23	
AMMONIA NITROGEN MG/L	0.6	. 1.4	1.4	1.6	1.4	1.6	•
NITRATE NITROGEN MG/L	0.1	0.6	0.8	0.6	0.7	0.7	
SULFATES AS SU4 MG/L	77	55	63	85	52	75	
TOTAL PHOS.AS PO4 MG/L	0.4	Ú <b>. 7</b>	0.6	5.6	0.8	0.7	
PHEMOLS PPB	2	4	4	5	7	1	
OILS & GREASES MG/L	1.4	0.6	1.9	2.3	2.1	1.0	
FLUORIDES MG/L	0.32	0.31	0.25	0.30	0.40	0.44	
COLIFORM M.P.N./100ML.	1800	240	350	240	28	33	
MPN OF E COLI/100ML	33	14.	6	7	3.	11	
STD. PLATE COUNT/ML	100000	78000	1450	5000	25000	300	
DETERGENTS MG/L	*0.00	*O.00	*0.00	*O.00	*0.00	0.10	
CHEMICAL OXYGEN DEMAND MG/L		11	13	34	10	9	

<sup>\*</sup> DENOTES DATA NOT AVAILABLE

	AVERAGE	MAXIMUM	MINIMUM	MEDIAN
DISSCLUED OXYGEN MG/L	9.4	13.7	4.8	8.9
BOD MG/L	0.7	1.7	0.3	0.6
HYDROGEN ION CONC. PH	7.9	8.3	7.5	. 8.4.0
ALKALINITY MG/L	144.	153.	126.	149.
TOTAL RESIDUE MG/L	244.	304.	220.	234.
INGITION LOSS MG/L	74.	110.	38.	68.
TURBICITY AS SID2 MG/L	1.	4.	1.	. 1.
TCTAL HARDNESS MG/L	168.	180.	154.	170.
CHLORIDES MG/L	1.	4.	0.	1.
AMMONIA NITROGEN MG/L	0.2	0.3	0.0	0.2
SULFATES AS SO4 MG/L	51.	. 70.	40.	52.
NITRATE NITROGEN MG/L	0.2	0.5	0.0	0.1
TOTAL PHOS.AS PO4 NG/L	0.1	0.3	0.0	0.1
PHENCLS PPB	3.	7.	0.	2.
CILS AND GREASES NG/L	1.5	3.6	04	0.9
COLIFORMS M.P.N./100ML	39.	350.	0.	0.
M.P.N. OF E COLI/100ML	2.	14.	0.	0.
STANDARD PLATE COUNT/ML	1404.	10000•	1.	10.

	AVERAGE	MAXIMUM	MINIMUM	MEDIAN
DISSOLVED OXYGEN MG/L	12.2	13.4	10.6	12.4
BCD MG/L	0.9	1.9	0.1	0.7
HYCROGEN ION CONC.PH	8.1	8.3	7.8	8.1
ALKALINITY MG/L	162.	190.	141.	158.
TCTAL RESIDUE MG/L	289.	384.	227.	298.
INGITION LCSS MG/L	93.	138.	54.	76.
	4.	.8	<b>i.</b> .	4.
TOTAL HARDNESS MG/L	199.	212.	184.	200.
CHLORIDES MG/L	2.	4.	C.	1.
AMMONIA NITROGEN MG/L	0.1	0.3	0.0	0.1
SULFATES AS SO4 MG/L	68.		5.7.	69.
NITRATE NITROGEN MG/L	0.2	0.4	0 • C	0.1
TOTAL PHOS.AS PO4 MG/L	0 • 2	0.4	0.0	0.1
PHENCLS PPB	3.	8.	C.	2.
OILS AND GREASES MG/L	1.4	2.7	0.7	_1.0
COLIFCRMS M.P.N./100ML	225.	920.	0.	79.
M.P.N. CF E COLI/100ML	12.	49.	0.	11.
STANDARD PLATE COUNT/ML	2907.	8000.	0.	1000.

	AVERAGE	MAXIMUM	MININUM	MED.I AN
DISSOLVED OXYGEN MG/L	12.1	13.9	10.8	12.1
BOD MG/L	1.2	3.4	0.1	0.9
HYDROGEN ION CONC. PH		8.3	8_0	8.1
ALKALINITY MG/L	166.	180.	150.	164.
TOTAL RESIDUE NG/L	325.	550.	240.	302.
INGITION LOSS MG/L	101.	202.	28.	90.
TURBICITY AS SID2 NG/L	10	57.	2.	4.
TOTAL HARDNESS NG/L	207.	220.	192.	206.
CHLORIDES MG/L	1.	2.	0.	1.
APMONIA NITROGEN NG/L	0.2	0.3	0.1	-0.2
SULFATES AS SO4 NG/L	69.	83.	55.	70•
NITRATE NITROGEN NG/L	0.2	0.4	0.0	0.3
TOTAL PHOS.AS PO4 MG/L	0.2	0.7	0.0	0.1
PHENOLS	4.	10.		3.
OILS AND GREASES NG/L	0.9	1.5	0.3	0.9
COLIFORMS M.P.N./100ML	285.	1600.	0.	79.
M.P.N. OF E COLI/100ML	57.	170.	0.	34.
STANDARD PLATE COUNTIME	7924.	55000.	10.	100.

B-25

	AVERAGE	MUMIXAM	MINIMUM	MEDIAN
DISSCLUED DXYGEN NG/L	12.3	15.5	10.2	13.0
BCD MG/L	0.6	1.3	0.1	0.6
HYDROGEN ION CCNCPH	8.0	8.5	7.0	8.1
ALKALINITY MG/L	163.	181.	112.	170.
TOTAL RESIDUE MG/L	286.	354.	232.	282
INGITION LOSS MG/L	74.	134.	32.	56
TURBIDITY AS SIDE NG/L	4	8	2	4,
TETAL HARDNESS MG/L	206.	212.	200.	206
CHLORIDES MG/L	2.	5.	0.	1.
AMMONIA NITROGEN MG/L	0.2	0.6	0.0	0.2
SULFATES AS SO4 MG/L	73.	96.	57.	73.
NITRATE NITROGEN MG/L	0.2	0.4	0.0	0.2
TOTAL PHOS.AS PO4 MG/L	0.2	0.7	0.0	0.1
PHENOLS	··· 4.	15.	1.	2
DILS AND GREASES MG/L	1.2	2.0	0.7	1.3
FLUORIDES MG/L	0.28	0.54	0.01	0.27
CCLIFCRMS M.P.N./100ML	409.	2400.+	10.	140.
M.P.N. CF E COLI/100ML	34.	170.	0	17.
STANDARD PLATE COLNI/ML	7408.	52000.	100	600

			<del></del>		
		AVERAGE	MAXIMUM	MINIMUM	MEDIAN
DISSOLVED OXYGEN	MG/L	11.6	14.7	9.5	11.1
BCC	MG/L	2.9	5.2	1.3	2.5
HYDROGEN ION CONC	- PH				
ALKALINITY MG/L		194.	280.	165.	176.
TOTAL RESIDUE	NG/L	297.	344.	242.	302.
INGITION LOSS MG/	L	103.	160.	62.	92.
TURBICITY AS SID2	MG/L	6.	12	1	<b>7.</b>
CHLORIDES	MG/L	5.	8.	1.	5.
APMONIA NITROGEN	MG/L	1.1	1.4	0.7	1.1
NITRATE NITROGEN	MG/L	0.3	0.6	0.0	0.3
TOTAL PHOS.AS PO4	MG/L	0.8	1.5	0.6	0.7
PHENOLS	PPB	5.	14.	1.	3.
OILS AND GREASES	MG/L	1.5	4.6	0.3	1.0
COLIFCRMS M.P.N./	100ML	39144.	180000.	23.	350.
M.P.N. OF E COLI/	100ML	1070.	2700.	0.	16.
STANDARD PLATE COL	NT/FL	217625.	950000.	<u> 4</u> 300.	100000.
DETERGENTS	MG/L.	0.30	0.50	0.20	0.20

	AVERAGE	MUMIXAM	MINIMUM	MEDIAN
DISSCLVED DXYGEN MG/L	11.1	13.6	8.7	10.0
BCC MG/L	2.2	3.2	0.9	2.5
HYDROGEN ION CONCPH	7.8	8.1	7.1	7.9
ALKALINITY MG/L	181.	211.	166.	176.
TUTAL RESIDUE MG/L	327.	386.	254.	318.
INGITION LOSS MG/L	99.	154.	34.	98.
TURBICITY AS SID2 MG/L	5.	9.	1.	4.
CILCRIDES MG/L	8.	29.	0.	6.
AMMONIA NITROGEN MG/L	1.9	2.3	1.4	1.9
NITRATE NITROGEN MG/L	0.6	1.1	0.1	0.5
TOTAL PHOS.AS PO4 MG/L	2.5	15.0	0.6	0.9
PHENCLS PPB	5.	12.	1.	4.
CILS AND GREASES MG/L	0.8	1.5	0.3	0.8
FLUCRIDES MG/L	0.28	0.48	0.02	0.28
CULIFORMS M.P.N./100ML	22794.	180000.	19.	2100.
M.P.N. OF E CCLI/100ML	2161.	14000.	4.	36.
STANDARD PLATE COUNT/ML	56178.	200000.	260C.	33000.

	AVERAGE	MAXIMUM	MINIMUM	MEDIAN
CISSULVED CXYGEN MG/L	10.9	12.7	8.8	10.8
BCC MG/L	2.5	4.9	1.1	2.4
HYDRUGEN ION CONG.,PH	7.9	8.2	7.6	8.0
ALKALINITY MG/L	181.	196.	168.	176.
TOTAL RESIDUE MG/L	334.	412.	273.	338.
INGITION LOSS MG/L	104.	134.	62.	108.
TUREIDITY AS SIO2 MG/L	7.	39.	1.	3.
CHLORIDES MG/L	10.	20.	4.	9.
AMMONIA NITROGEN MG/L	1.3	1.6	0.9	1.4
NITRATE NITROGEN MG/L	0.4	0.7	0.C	0.6
TCTAL PHOS.AS PO4 MG/L	0.9	1.3	0.6	8.0
PHENCLS . PPB	7.	13.	1.	6.
CILS AND GREASES MG/L	0.6	1.0	0.3	0.4
CCLIFCRMS M.P.N./100ML	4274.	18000.	11•	170.
M.P.N. OF E CCLI/100ML	224.	1500.	C.	14.
STANDARD PLATE COUNT/ML	43933.	300000.	10C.	12000.

	AVERAGE	MAXIMUM	MINIMUM	MEDIAN
DISSCLVED CXYGEN MG/L	. 8.9	11.5	7.5	8.5
BCD MG/L	1.7	2.8	1.0	1.6
HYDROGEN ION CONC.,PH	8.0	8.2	7.8	0.8
ALKALINITY MG/L	185.	197.	170.	186.
TUTAL RESIDUE MG/1	367.	454.	293.	354.
INGITION LCSS MG/L	111.	146.	76.	112.
TURBICITY AS SIO2 MG/L	. 2.	5.	1.	2.
CHLORIDES MG/1	. 18.	33.	2.	19.
AMMONIA NITROGEN MG/1	1.5	2.2	0.6	1.5
NITRATE NITROGEN MOZI	0.3	0.6	0.1	0.3
TOTAL PHOS.AS PO4 MG/I	0.9	1.5	0.•3	8.0
PHENOLS PPE	6.	12.	C.	6.
CILS AND GREASES MC/I	1.4	2.5	0.3	1.0
CULIFORMS M.P.N./100MI	482.	1800.+	0.	46.
M.P.N. OF E COLI/100MI	27.	220.	C.	4.
STANDARD PLATE COUNT/MI	299822.	1100000.	1300.	190000.

		AVERAGE	MAXIMUM	MINIMUM	MEDIAN
DISSOLVED CXYGEN	MG/L	8.9	13.1	5.7	9.5
BCC	G/L	1.9	3.9	1.0	1.8
HYDROGEN ION CONC.	• PH	8.0	8.3	7.7	8.0
ALKALINITY .MG/L		183.	194.	173.	184.
TOTAL RESIDUE	MG/L	353.	408.	296.	360.
INGITION LOSS MO/L		107.	138.	84.	100.
TURBICITY AS STO2	MG/L	4.	8.	1.	3.
CHLORIDES	MG/L	14.	25.	<b>G</b> •	17.
APMONIA NITROGEN	MG/L	1.5	2.4	0.7	1.5
NITRATE NITROGEN	MG/L	0.5	0.7	0.1	0.6
TOTAL PHOS.AS PO4	MG/L	0.7	1.0	0.5	0.6
PPENULS	PPB	4.	12.	G.	4.
CILS AND GREASES	MG/L	1.7	3.4	0.3	1.5
CULIFORMS M.P.N./1	OOML	239.	1800.+	5.	24.
Mapana OF E COLIZI	00 M L	9.	22.	2.	8.
STÄNDARD PLATE COLN	T/ML	84933.	620000.	300.	12000.

		AVERAGE	MAXIMUM	MINIŅUM	MEDIAN
DISSOLVED DXYGEN	MG/L	7.2	9.2	5.6	6,5
ecc	MG/L	1.8	3.9	0.4	1.3
HYDROGEN ION CONC	<b>.</b> PH	8 • 0	8.4	7.7	8,0
ALKALINITY MG/L	•	188.	203.	169.	188.
TOTAL RESIDUE	MG/L	376.	464.	306.	374.
INGITION LOSS MG	/L	100.	134.	54.	98.
TUREICITY AS SIO	2 MG/L	4.	10.	1.	3.
TOTAL HARDNESS	MG/L	220.	238.	204.	222.
CHLORIDES	MG/L	27.	48.	13.	26.
APMONIA NITROGEN	MG/L	1.4	1.9	0.6	1.4
SULFATES AS SU4	MG/L	73.	89.	52.	75.
NITRATE NITROGEN	MG/L	0.5	0.8	0.1	0.6
TOTAL PHUS.AS PO-	4 MG/L	1.3	5.6	0.4	0.7
PHENCLS	РРВ	5.	11.	1.	4.
DILS AND GREASES	MG/L	1.4	2.3	0.6	1,4
FLUCRIDES MG/L		0.29	0.44	0.01	0.30
CELIFERMS M.P.N.	/100ML	300.	1800.	0.	33•
M.P.N. OF E COLI	/1COML	8.	33.	<b>C</b> •	7.
STANDARD PLATE CO		127794.	570000.	300.	25000.

#### APPENDIX C

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Lab. Sample No. S-151	<u>.                                    </u>		
Type of Sample Water			
Source of Sample NS	SR at 105 Street Bri	ldge	
Date Sampled June	17/70	Date Received June 17/70	
Results of Requested	Tests: Herbicides	and Pesticides - 4000 ml extra	cted
2,4-D Acid	0.06 µg/1	o,p-DDT	N.D. /ug/1
2,4,5-T Acid	N.D. /ug/1	p,p-DDT	N.D. /ug/l
Silvex Acid	N.D. ug/1	DDVP	N.D. /ug/l
MCPA Acid	N.D. ug/l	Ronnel	N.D. /ug/l
Tordon Acid	N.D. ug/1	Parathion	N.D. /ug/l
Dicamba Acid	N.D. ug/l	Diazinon	N.D. jug/1
Methoxychlor	N.D. ug/1	Malathion	N.D. ug/l
Lindane	N.D. ug/1	2,4-D ethyl ester	N.D. /ug/l
Endrin	N.D. ug/1	2,4-D isopropyl ester	N.D. /ug/l
Dieldrin	N.D. ug/l	2,4-D n-butyl ester	N.D. /ug/1
Heptachlor	N.D. /ug/l	2,4-D iso-butyl ester	N.D. ug/l
Heptachlor E <sub>l</sub> oxide	N.D. /ug/1	2,4,5-T methyl ester	N.D. /ug/1
Aldrin	N.D. /ug/1	2,4,5-T n-amyl ester	<u>N.D.</u> ug/l
DDE	N.D. /ug/l	2,4,5-T isopropyl ester	N.D. /ug/l
D <b>DD</b>	N.D. /ug/1	2,4,5-T n-butyl ester	N.D. /ug/l
Trace is Defined as a	Quantity less than	0.01/ug/l, But still detectable	le.
N.D Not detected.			
REMARKS:			
***************************************	·		
			<del></del>
Date: August 19, 19	70 Certi	fied	
Form LHP-0010			

Lab. Sample No. S-152			
Type of Sample Water			
Source of Sample NSR a	t Vinca		
Date Sampled June 17	'/70 I	Pate Received June 17/70	
Results of Requested Te	sts: Herbicides a	nd Pesticides - 4000 ml. extra	acted
2,4-D Acid	0.08 ug/1	o,p-DDT	N.D. ug/
2,4,5-T Acid	N.D. ug/1	p,p-DDT	N.D. /ug/]
Silvex Acid	N.D. /ug/1	DDVP	N.D. ug/
MCFA Acid	N.D. /ug/1	Ronnel	N.D. /ug/]
Tordon Acid	N.D. ug/1	Parathion	N.D. /ug/]
Dicamba Acid	N.Dug/1	Diazinon	N.Dug/1
Methoxychlor	N.D. /ug/1	Malathion	. N.D. /ug/1
Lindane	N.D. /ug/l	2,4-D ethyl ester	N.D. ug/1
Endrin		2,4-D isopropyl ester	
Dieldrin	N.D. /ug/l	2,4-D n-butyl ester	N.D. /ug/1
Heptachlor	N.D. ug/1	2,4-D iso-butyl ester	N.D. ug/1
Heptachlor Epoxide	N.D. /ug/l	2,4,5-T methyl ester	N.Dug/1
Aldrin	ug/l	2,4,5-T n-amyl ester	N.Dug/1
DDE		2,4,5-T isopropyl ester	N.D. /ug/l
DDD	N.D. /ug/l	2,4,5-T n-butyl ester	N.D. ug/1
Trace is Defined as a qu	antity less than 0	.01/ug/1, But still detectable	
N.D Not detected.			
REMARKS:			
-			
Date: August 19, 1970	Certifie	d T	
Form LHP-0010			

Source of Sample NS		Date Received June 18/70	
		2400 MODELVER	
Results of Requested		and Pesticides - 4000 ml. ext	racted
2,4-D Acid	0.2ug/1	o,p-DDT	N.D. ug/
2,4,5-T Acid	Trace ug/l	p,p-DDT	
Silvex Acid	N.D. ug/l	DDVP	N.D. /ug/.
MCPA Acid	N.D. /ug/l	Ronnel	N.D. ug/
ordon Acid	N.D. ug/l	Parathion	N.D. /ug/
icamba Acid	N.D. ug/l	Diazinon	N.D. /ug/
leth <b>oxychlor</b>	N.D. ug/l	Malathion	N.D. ug/
ind <b>ane</b>	N.D. ug/l	2,4-D ethyl ester	N.D. ug/
ind <b>rin</b> ,	N.D. ug/l	2,4-D isopropyl ester	N.D. /ug/l
ieldrin	N.D. ug/l	2,4-D n-butyl ester	N.D. /ug/
dep <b>tachlor</b>	N.D. ug/1	2,4-D iso-butyl ester	N.D. ug/
eptachlor Epoxide	N.D. ug/1	2,4,5-T methyl ester	N.D. /ug/]
ldrin	N.D. ug/1	2,4,5-T n-amyl ester	N.D. /ug/1
DE	N.D. ug/l	2,4,5-T isopropyl ester	N.D. ug/
DD	N.D. ug/l	2,4,5-T n-butyl ester	N.D. ug/
race is Defined as a	•	0.01/ug/1, But still detectable	,
I.D Not detected.		,	
EMARKS:	•		

Lab. Sample No. S-154			•
Type of Sample Water			
Source of Sample Vinca	Ferry, NSR	·	
Date Sampled June 23,	1970	Date Received June 23, 1970	
Results of Requested Tes	sts: 4000 ml.	extracted	
2,4-D Acid	0.2 ug/1	o,p-DDT	N.D. ug/l
2,4,5-T Acid	0.09 ug/1	p,p-DDT	N.D. /ug/l
Silvex Acid	N.D. ug/1	DD <b>V</b> P	N.D. /ug/1
MCPA Acid	0.6 ug/1	Ronnel	N.D. /ug/1
Tordon Acid	N.D. ug/l	Parathion	N.D. ug/1
Dicamba Acid	Trace ug/l	Diazinon	N.D. /ug/1
Methoxychlor	N.D. ug/1	Malathion	N.D. ug/1
Lindane	N.D. ug/l	2,4-D ethyl ester	N.D. ug/1
Endrin	N.D. ug/1	2,4-D isopropyl ester	N.D. /ug/1
Dieldrin	N.D. ug/1	2,4-D n-butyl ester	N.D. ug/1
Heptachlor	N.D. ug/1	2,4-D iso-butyl ester	N.D. /ug/l
Heptachlor Eroxide	N.D. ug/1	2,4,5-T methyl ester	N.D. /ug/l
Aldrin	N.D. ug/1	2,4,5-T n-amyl ester	N.Dug/1
DDE	N.D. ug/l	2,4,5-T isopropyl ester	N.D. ug/l
DDD	N.D. ug/l	2,4,5-T n-butyl ester	N.D. /ug/1
Trace is Defined as a qu	antity less than	0.01/ug/l, But still detectable	,
N.D Not detected.			
REMARKS:			
Date: August 19, 1970	Certif	fied	
Form LHP-0010			

ATTENTION: E. Kupchanko
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Lab. Sample No. S-155						
Type of Sample Water						
Source of Sample NSR - 1	105 Street Bridge					
Date Sampled June 30	), 1970 Dat	e Received June 30, 1970				
Results of Requested Tests: Herbicides and Pesticides - 4000 ml. extracted						
2,4-D Acid	N.D. ug/l	o,p-DDT	N.Dug/l			
2,4,5-T Acid	ug/l	p,p-DDT	N.Dug/1			
Silvex Acid	N.Dug/l	DDVP	N.D. /ug/1			
MCFA Acid	N.D. ug/l	Ronnel	N.Dug/1			
Tordon Acid	N.D. ug/1	Parathion	N.Dug/1			
Dicamba Acid	N.D. ug/l	Diazinon	N.Dug/1			
Methoxychlor	N.D. ug/1	Malathion	N.D. /ug/1			
Lindane	N.D. ug/l	2,4-D ethyl ester	N.Dug/1			
Endrin	N.D. ug/1	2,4-D isopropyl ester	N.Dug/1			
Dieldrin	N.D. ug/l	2, h-D n-butyl ester	N.Dug/l			
Heptachlor	N.D. ug/l	2,4-D iso-butyl ester	N.Dug/1			
Heptachlor Eroxide	N.D. ug/l	2,4,5-T methyl ester	N.D. /ug/1			
Aldrin	$N_{\bullet}D_{\bullet}$ ug/1	2,4,5-T n-amyl ester	N.Dug/l			
DDE	N.D. ug/l	2,4,5-T isopropyl ester	N.D. /ug/l			
מממ	N.D. ug/l	2,4,5-T n-butyl ester	N.D. /ug/1.			
Trace is Defined as a quantity less than 0.01/ug/1, But still detectable.						
N.D Not detected.						
REMARKS:						
			· · · · · · · · · · · · · · · · · · ·			
Date: August 19, 1970	Certifie	d				

- c6 -

Source of Sample NSI  Date Sampled Jur	ne 30, 1970	Date Received June 30, 1970	
		nd Herbicides - 4000 ml. extr	acted ( ) lizzi
2,4-D Acid	0.4 /ug/l	o,p-DDT	N.D. ug/
2,4,5-T Acid	0.1 /ug/1	p,p-DDT	N.D. /ug/]
Silvex Acid	N.Dug/1	DDVP	
MCFA Acid		Ronnel	
Tordon Acid			
Dicamba Acid		Parathion	N.D. /ug/]
	N.D.	Diazinon	N.D.
Methoxychlor	N.D. jug/1	Malathion	N.D. /ug/1
Lindane	N.D. ug/1	2,4-D ethyl ester	N.D.
Endrin	N.Dug/1	2,4-D isopropyl ester	N.D. /ug/1
Dieldrin	N.D. ug/l	2,4-D n-butyl ester	N.D. ug/
Heptachlor	N.D. ug/l	2,4-D iso-butyl ester	N.D. /ug/1
Heptachlor Epoxide	N.D. ug/1	2,4,5-T methyl ester	N.D. ug/
Aldrin	N.D. ug/1	2,4,5-T n-amyl ester	N.D. ug/
DDE	N.D. ug/l	2,4,5-T isopropyl ester	N.D. ug/1
DDD	N.D. ug/l	2,4,5-T n-butyl ester	N.D. ug/]
Trace is Defined as a	quantity less than	0.01/ug/1, But still detectable	,
N.D Not detected.		<b>/</b>	
REMARKS:			
			,

ATTENTION: E. KUPCHANKO

Lab. Sample No. 8-187	)		
Type of Sample water			
Source of Sample NSR	- Vinca		
Date Sampled July	15/70 I	Date Received July 15/70	
Results of Requested 1	'ests: <u>Herbicides a</u>	and Pesticides - 4000 ml. extra	ıcted
2,4-D Acid	N.D. ug/1	o,p-DDT	N.D. /ug/l
2,4,5-T Acid	N.D. ug/l	p,p-DDT	N.D. /ug/l
Silvex Acid	N.Dug/l	DDVP	N.D. /ug/1
MCPA Acid	0.1 ug/1	Ronnel	N.D. ug/1
Tordon Acid	N.D. ug/1	Parathion	N.D. /ug/1
Dicamba Acid	N.D. ug/l	Diazinon	N.D. /ug/1
Methoxychlor	N.D. ug/l	Malathion	N.D. /ug/l
Lindane	N.D. ug/1	2,4-D ethyl ester	N.D. /ug/1
End <b>rin</b>	N.D. ug/1	2,4-D isopropyl ester	N.Dug/1
Dieldrin	N.D. ug/l	2,4-D n-butyl ester	N.D. ug/l
Hept <b>achl</b> o <b>r</b>	N.D. ug/1	2,4-D iso-butyl ester	N.D. ug/l
Heptachlor Epoxide	$\frac{\text{N.D.}}{\text{/}} \text{ug/1}$	2,4,5-T methyl ester	N.D. ug/1
Aldrin	N.D. ug/1	2,4,5-T n-amyl ester	N.D. /ug/l
DDE	N.D. ug/1	2,4,5-T isopropyl ester	N.D. /ug/l
<b>ס</b> מת	N.D. ug/l	2,4,5-T n-butyl ester	N.D. ug/l
Trace is Defined as a	quantity less than (	0.01/ug/l, But still detectable	₽•
N.D Not detected.			
REMARKS:			
			<del></del>
			<del></del>
Date: August 19, 197	Certif	ied	

ATTENTION:	E.	Kupchanko	
*** * *** * * * * * * * * * * * * * * *		-	

Lab. Sample No. S-191				
Type of Sample water			·	
Source of Sample White	mud Creek above Rai	nbow Valley		
Date Sampled July	2 <b>4/7</b> 0 I	Date Received July 24/70		· · · · · · · · · · · · · · · · · · ·
Results of Requested Te	sts: Pesticides	and Herbicides - 4000 ml. ext	racted	<del></del>
2,4-D Acid	N.D. ug/1	o,p-DDT	N.D.	ug/l
2,4,5-T Acid	N.D. ug/1	p,p-DDT	N.D.	ug/1
Silvex Acid	N.D. /ug/1	DD <b>V</b> P	N.D.	ug/1
MCFA Acid	N.D. /ug/l	Ronnel	N.D.	ug/1
Tordon Acid	N.D. ug/1	Parathion	N.D.	ug/1
Dicamba Acid	N.D. ug/1	Diazinon	N.D.	ug/1
Methoxychlor	N.D. /ug/1	Malathion	N.D.	ug/1
Lindane	N.D. ug/1	2,4-D ethyl ester	N.D.	ug/1
Endrin	N.D. ug/1	2,4-D isopropyl ester	N.D.	ug/1
Dieldrin	N.D. /ug/1	2,4-D n-butyl ester	N.D.	ug/1
Heptachlor	N.D. ug/1	2,4-D iso-butyl ester	N.D.	ug/1
Heptachlor $E_I$ oxide	N.D. ug/l	2,4,5-T methyl ester	N.D.	ug/1
Aldrin	N.D. /ug/1	2,4,5-T n-amyl ester	N.D.	ug/1
DDE	N.D. ug/l	2,4,5-T isopropyl ester	N.D.	ug/1
מסמ	N.D. /ug/1	2,4,5-T n-butyl ester	N.D.	•
Trace is Defined as a q	uantity less than (	0.01/ug/l, But still detectabl		7
N.D Not detected.		·		
REMARKS:				
	·			
Date: August 19/70	Certifi	ed Th		
Form LHP-0010			_	

ate Sampled Ju	1 <del>y</del> 24/70	Date Received July 24/70	
esults of Requested	Tests: Pesticides	and Herbicides - 4000 ml. extra	acted
,4-D Acid	0.09 ug/1	o,p-DDT	N.D.
,4,5-T Acid	N.Dug/l	p,p-DDT	N.D.
ilvex Acid	N.D. ug/1	DDVP	N.D.
CFA Acid	0.6 ug/1	Ronnel	N.D. ug
ordon Acid	N.D. ug/1	Parathion	N.D.
icamba Acid	N.D. ug/1	Diazinon	N.D.
ethoxychlor	N.D. ug/1	Malathion	N.D.
indane	N.D. ug/1	2,4-D ethyl ester	N.D.
ndrin	N.D. ug/1	2,4-D isopropyl ester	N.D.
ieldrin	N.D. ug/1	2,4-D n-butyl ester	N.D.
ep <b>tachlor</b>	N.D. ug/1	2,4-D iso-butyl ester	N.D.
eptachlor Epoxide	N.D. ug/1	2,4,5-T methyl ester	N.D.
ldrin	N.Dug/1	2,4,5-T n-amyl ester	N.D.
DE	N.D. ug/l	2,4,5-T isopropyl ester	N.D.
מס	N.D. ug/1	2,4,5-T n-butyl ester	N.D.
race is Defined as a	quantity less than	0.01/ug/l, But still detectable	e.
.D Not detected.			
EMARKS:			

ATTENTION:	$\mathbf{E}_{ullet}$	Kupchanko	_

C 10 -

Lab. Sample No. S-193			
Type of Sample water			
Source of Sample N.S.R.	at Fort Saskatch	newan	
Date Sampled July 2	24/70	Date Received July 24/70	
Results of Requested Tes	ts: Herbicides a	and Pesticides - 4000 ml. extra	acted
2,4-D Acid	N.Dug/1	o,p-DDT	N.D. ug/1
2,4,5-T Acid	N.D. ug/1	p,p-DDT	N.D. /ug/1
Silvex Acid	N.D. ug/1	DD <b>V</b> P	N.D. /ug/1
MCFA Acid	N.D. ug/l	Ronnel	N.D. ug/1
Tordon Acid	N.D. ug/1	Parathion	N.D. /ug/1
Dicamba Acid	N.D. ug/1	Diazinon	N.D. /ug/1
Methoxychlor	N.D. ug/1	Malathion	N.Dug/1
Lindane	N.D. ug/1	2,4-D ethyl ester	N.D. ug/l
Endrin	N.Dug/1	2,4-D isopropyl ester	N.D. /ug/1
Dieldrin	N.D. ug/1	2,4-D n-butyl ester	N.Dug/l
Heptachlor	N.D. ug/1	2,4-D iso-butyl ester	N.Dug/1
Hertachlor Epoxide	N.D. ug/1	2,4,5-T methyl ester	N.Dug/1
Aldrin	N.D. ug/1	2,4,5-T n-amyl ester	N.D. ug/1
DDE	N.D. ug/1	2,4,5-T isopropyl ester	N.D. /ug/1
ממת	N.D. ug/l	2,4,5-T n-butyl ester	N.Dug/l
Trace is Defined as a qu	•	0.01/ug/1, But still detectabl	<del></del>
N.D Not detected.		,	
REMARKS:			
			THE RESERVE OF THE PARTY OF THE
Date: August 19, 1970	Certif:	ied 💮	
Form LHP-0010	-		

Source of Sample N.S.			
Date Sampled July		Date Received July 24/70	
	Tests: <u>Pesticides</u>	and Herbicides - 4000 ml.	
2.4-D Acid	N.D. ug/1	o,p-DDT	N.Dug/]
2,4,5-T Acid	N.D. ug/l	p,p-DDT	N.D. /ug/1
Silvex Acid	N.D. ug/1	DDVP	N.Dug/1
MCFA Acid	0.2 /ug/1	Ronnel	N.D. /ug/1
forden Acid	N.D. ug/1	Parathion	N.D. ug/1
Dicamb <b>a Acid</b>	N.D. ug/1	Diazinon	N.Dug/1
4cthoxychlor	N.D. ug/l	Malathion	N.Dug/]
Lindane	N.D. ug/1	2,4-D ethyl ester	N.D. ug/i
En <b>drin</b>	N.D. ug/1	2,4-D isopropyl ester	N.Dug/1
Die <b>ldrin</b>	N.D. ug/1	2,4-D n-butyl ester	N.D. /ug/l
lep <b>tachlor</b>	N.D. ug/l	2,4-D iso-butyl ester	N.D
ieptachlor E <sub>l</sub> oxide	N.D. ug/1	2,4,5-T methyl ester	N.D. /ug/l
Aldrin	N.D. ug/1	2,4,5-T n-amyl ester	N.D. ug/l
DDF.	•	2,4,5-T isopropyl ester	N.D. ug/1
<b>מ</b> סכ	N.D. /ug/l	2,4,5-T n-butyl ester	N.Dug/3
Frace is Defined as a	quantity less than	0.01/ug/l, But still detectable	•
N.D Not detected.		·	
REMARKS:			

ATTENTION:	E.	Kupchanko	
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- C 12 -

ate Sampled Ju	Ly 24/70 1	Date Received July 24/70	
esults of Requested	Tests: Pesticides	and Herbicides - 4000 ml. ext	racted
,4-D Acid	N.D. ug/1	o,p-DDT	N.D. ug/
,4,5-T Acid	N.D. /ug/1	p,p-DDT	N.D. ug/
ilvex Acid	N.Dug/1	DD <b>V</b> P	N.D. /ug/
CPA Acid	N.D. ug/1	Ronnel	N.D. ug/
ordon Acid	N.D. ug/1	Parathion	N.D. ug/
licamba Acid	N.D. ug/l	Diazinon	N.D. ug/
ethoxychlor	N.D. ug/1	Malathion	N.Dug/
indane	N.D. ug/1	2,4-D ethyl ester	N.D. ug/
indrin	N.D. ug/1	2,4-D isopropyl ester	N.D. ug/
rieldrin	N.D. ug/l	2,4-D n-butyl ester	N.D. ug/
ertachlor	N.D. ug/l	2,4-D iso-butyl ester	N.D. jug/
ertachlor Eloxide	N.D. ug/l	2,4,5-T methyl ester	N.D. ug/
ldrin	N.D. ug/1	2,4,5-T n-amyl ester	N.D. ug/
DE	N.D. ug/1	2,4,5-T isopropyl ester	N.D. ug/
מת	N.D. ug/l	2,4,5-T n-butyl ester	N.D. ug/
race is Defined as a	quantity less than	0.01/ug/1, But still detectabl	,
J.D Not detected.			
EMARKS:			

ATTENTION: E. Kupchanko

ype of Sample water		d dan	
ource of Sample N.S.	00.45		
,4-D Acid		nd Herbicides - 4000 ml. extra	
,4,5-T Acid	N.D. /ug/1	o,p-DDT	N.D. /ug/
•	N.D. ug/1	p,p-DDT	N.D. /ug/
ilvex Acid	N.D. /ug/1	DDVP	N.Dug/.
CPA Acid	0.3 /ug/1	Ronnel	N.D. ug/
ordon Acid	N.D. ug/1	Parathion	N.D. ug/
icamba Acid	$N.D.$ $\mu g/1$	Diazinon	N.D. /ug/
ethoxych <b>lor</b>	N.D. /ug/l	Malathion	N.D. ug/
ind <b>ane</b>	N.D. ug/l	2,4-D ethyl ester	N.D. ug/
ndrin	N.D. ug/l	2,4-D isopropyl ester	N.D. /ug/
ieldrin	N.D. ug/l	2,4-D n-butyl ester	N.Dug/
ep <b>tachlor</b>	N.D. ug/l	2,4-D iso-butyl ester	N.D. ug/
eptachlor Epoxide	N.D. ug/1	2,4,5-T methyl ester	N.Dug/:
ldrin	N.D. ug/l	2,4,5-T n-amyl ester	N.D. ug/
DE	•	2,4,5-T isopropyl ester	N.Dug/:
DD	•	2,4,5-T n-butyl ester	N.D. ug/
race is Defined as a	•	0.01/ug/1, But still detectable	, e•
D Not detected.		,	
EMARKS:			
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	,		
,			

C 14 -

ate Sampled July 28/	<b>'</b> 70	Date Received July 28/70	
Results of Requested	Tests:		
2,4-D Acid	N.D. ug/1	o,p-DDT	N.D. ug/
.4,5-T Acid	N.D. ug/l	p,p-DDT	N.D. /ug/]
Cilvex Acid	N.D. /ug/1	DDVP	N.D. /ug/1
MCFA Acid	N.Dug/l	Ronnel	<b>N.D.</b> _ug/]
ordon Acid	N.Dug/l	Parathion	N.D. /ug/1
Dicamba Acid	N.D. ug/1	Diazinon	N.D. jug/l
lethoxychlor	N.D. ug/1	Malathion	N.D. /ug/]
Lindane	N.D. /ug/l	2,4-D ethyl ester	N.D. /ug/1
ndrin	N.D. ug/1	2,4-D isopropyl ester	N.D. /ug/]
Dieldrin		2,4-D n-butyl ester	N.D. /ug/]
leptachlor	N.D. ug/l	2,4-D iso-butyl ester	N.D. ug/l
ieptachlor Epoxide	N.D. ug/l	2,4,5-T methyl ester	N.D. /ug/l
aldrin	N.Dug/l	2,4,5-T n-amyl ester	N.D. ug/1
DE		2,4,5-T isopropyl ester	N.D. ug/1
COC	N.D. ug/1	2,4,5-T n-butyl ester	N.Dug/1
race is Defined as a	quantity less than	0.01/ug/l, But still detectable	,
N.D Not detected.			
EMARKS:			
	·		

ATTENTION:	E. Kunchanko
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Date Sampled July 2	28/70	Date Received July 28/70	
Results of Requested	Tests:		
2.4-D Acid	N.D. /ug/l	o,p-DDT	N.D. ug
2,4,5-T Acid	N.D. ug/l	p,p-DDT	<b>N.D.</b> /ug/3
Silvex Acid	N.D. /ug/l	DD <b>V</b> P	N.D. ug/
MCFA Acid	N.D. /ug/l	Ronnel	N.D. ug/1
fordon Acid	N.D. ug/l	Parathion	N.D. aug/]
Dicamba Acid	N.D. ug/l	Diazinon	<b>N.D.</b> etc/1
Methoxychlor	N.D. /ug/l	Malathion	N.D. /ug/]
Lindane	N.D. ug/1	2,4-D ethyl ester	N.D. jug/
Endrin	N.D. ug/l	2,4-D isopropyl ester	<b>N.D.</b> /ug/3
Dieldrin	N.D. /ug/l	2,4-D n-butyl ester	N.D. /ug/]
Hep <b>tachlor</b>	N.D. ug/l	2,4-D iso-butyl ester	N.D. ug/
Heptachlor E <sub>l</sub> oxide	N.D. ug/l	2,4,5-T methyl ester	<b>N.D.</b> ug/l
Aldrin	N.D. ug/l	2,4,5-T n-amyl ester	N.D. ug/1
DDE	·	2,4,5-T isopropyl ester	
<b>d</b> dd		2,4,5-T n-butyl ester	
Trace is Defined as a	quantity less than	0.01/ug/1, But still detectable	<b>}</b> •
N.D Not detected.			
REMARKS:			·

lab. Sample No. S - ]	199		
Type of Sample Water	•		
Source of Sample N.S.F	R. @ Vinca		
Date SampledJuly 28/	/70	Date ReceivedInly 28/70	
Results of Requested	Tests:		· <del></del>
2,4-D Acid	N.D. /ug/l	o,p-DDT	N.D. /ug/1
2,4,5-T Acid	N.D. /ug/l	p,p-DDT	
Silvex Acid	N.D. /ug/l	DDVP	
MCPA Acid	N_Dug/l	Ronnel Ronnel	N.D. /ug/l
Tordon Acid	_N_Dug/1	Parathion	
Dicamba Acid	N.D. /ug/l	Diazinon	
Methoxychlor	N.Dug/l	Malathion	N.D. /ug/1
Lindane		2,4-D ethyl ester	N.D. yug/l
Endrin	N.D. /ug/1	2,4-D isopropyl ester	N.D. /ug/l
Dieldrin	$N.D.$ $\mu$ g/1	2,4-D n-butyl ester	
Her tachlor	N.D. /ug/l	2,4-D iso-butyl ester	N.D. jug/3
Hertachlor Eloxide		2,4,5-T methyl ester	
Aldrin	N.Dug/l	2,4,5-T n-amyl ester	
DDĿ		2,4,5-T isopropyl ester	ug/1
ממנו	N.D. ug/l	2,4,5-T n-butyl ester	
Trace is Defined as a	quantity less than	0.01/ug/l, But still detectable	· · · · · · · · · · · · · · · · · · ·
N.D Not detected.			
EMARKS:			
1			
Date: August 31, 1	970 Certif	Fied	
Form LHP-0010			

- 1/2 M LIH =00.

iah. Sample No. S - 200 Type of Sample Water Source of Sample N.S.R. @ Ft. Sask. Bridge Date Sampled July 28/70 Date Received July 28/70 Results of Requested Tests: 2.4.D Acid N.D. /ug/1 o,p-DDT N.D. /ug 1 1,4,5-T Acid N.D. ug/l p,p-DDT N.B. /ug/1 Silvex Acid N.D. /ug/l DDVP <u>N.D.</u> yug/∴ MCFA Acid <u>N.D.</u>ug/l Ronnel <u>N.D.</u>ug/l Torden Acid  $N.D. \mu g/1$ Parathion N.D. aug/l Dicamba Acid N.D. \_ug/l Diazinon N.D. 22-77 Mathemychlor Malathion N.D. ug/l N.D. /ug/1 Lindane \_\_\_\_N.D.\_\_\_ug/l 2,4-D ethyl ester **N.D.** \_ug/\_ Endrin  $N.D. \mu g/1$ 2,4-D isopropyl ester N.D. /ug/l N.D.\_\_ug/1 Dieldrin 2,4-D n-butyl ester N.D. ng/1 Heptachlor N.D. \_ug/l 2,4-D iso-butyl ester **N.D.** ug/i N.D. \_ug/l 2,4,5-T methyl ester Heptachlor Ejoxide **N.D.** ug/l N.D. /ug/1 2,4,5-T n-amyl ester N.D. /ug/1 Aldrin N.D. ug/l 2,4,5-T isopropyl ester N.D. ug/1 DDL 2,4,5-T n-butyl ester DDD N.D. \_ug/l Trace is Defined as a quantity less than 0.01/ug/l, But still detectable. N.D. - Not detected. REMARKS: Certified Date: August 31, 1970

Form LHP-0010

ATTENTION:	E.	E.	Kupchanko	
	-	_		

- C 1.8 -

Lab. Sample No. 7	LP - 41			
Type of Sample:	Water		Ti	me:
Source of Sample:	Storm Sewer	Emily Murphy F	ark	
Date Sampled:	April 1/71		Date Receiv	ed: April 1/71
ACIDS				ESTERS
2, 4-D	E	ndrin		2, 4-D Methyl
2,4,5-T	М	ethoxychlor		Ethyl
Silvex	0	, P-DDT		iso-Propyl
MCPA	P	, P-DDT		n-Propyl
Tordon				iso-Butyl
Dicamba	<u>P</u> I	HOSPHOROUS PESTI	CIDES	n-Butyl
	Ro	onnel	<del></del>	n-Amyl
CHLORINATED PESTICI	DES Pa	arathion	<del></del>	2,4,5-T Methyl
Lindane	Me	ethyl Parathion		Ethyl
Heptachlor	Ma	alathion		iso-Propyl
Aldrin	D	iazinon	T	n-Propyl
Heptachlor Epoxide	E	thion		iso-Butyl
DDE	Me	ethyl Trithion		n-Butyl
DDD	Ph	norate		n-Amyl
Dieldrin		PCB		
Trace is defined as	a quantity le	ess than 0.1/ug/1	, but still	detectable. All concentra
tions are in ug/l o	f sample.			
REMARKS:05 ug/	1 Cl of unkno	wn phosphorous (	l peak) plus	2 unknown trace
peaks -	2 unknown tra	ce peaks of Chlo	rinated Pest	icide.
0.3 ug	/1 Cl of unkno	wn Esters (2 pea	ks)	
Date: May 4,	19/1		Certified:	THE STATE OF THE S
Form THP_0010			•	

ATTENTION:	$\mathbf{E}_{ullet}$	$\mathbf{E}_{ullet}$	Kupchanko	
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Type of Sample: Wate	r	Time:
Source of Sample:	Storm Sewer Whitemud Ro	ad
Date Sampled: Apri	1 1/71	Date Received: April 1/71
ACIDS		ESTERS
2, 4-D	Endrin	2, 4-D Methyl
2,4,5-T	Methoxychlor	
Silvex	-	Ethyl
	O, P-DDT	iso-Propyl
MCPA	P, P-DDT	n-Propyl
Tordon .		iso-Butyl
Dicamba	PHOSPHOROUS PESTIO	CIDES n-Butyl
	Ronnel	n-Amyl
CHLORINATED PESTICIDES	Parathion	2,4,5-T Methyl
Lindane	Methyl Parathion	Ethyl
Heptachlor	Malathion	iso-Propyl
Aldrin	Diazinon	n-Propyl
Heptachlor Epoxide	Ethion	iso-Butyl
DDE	Methyl Trithion	n-Butyl
םמם	Phorate	T n-Amyl
Dieldrin	PCB	
Trace is defined as a cus	ntity less than 0 1 Aug/1	l, but still detectable. All c
		., 540 5422 400004020, 1122 0
tions are in mg/l of samp	Tr.	
REMARKS: 8 phosphorous u	nknown trace peaks	
.037 Aug/1 61 of u	nknown Chlorinated Pesti	cide plus 4 trace peaks
•	nown Esters (6 peaks)	
		&

	ATTENTION:	E.	E.	Kupchanko		
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Lab. Sample No. 71	P - 40					
Type of Sample:	Water		·	_ Time: _	<del></del>	
Source of Sample: _	Storm Sewe	r South end of (	Quenel	Bridge		
Date Sampled:	April 1/71		Date R	eceived: _	April 1	/71
ACIDS				EST	TERS	
2, 4-D	En	drin		2,	4-D Methyl	
2,4,5-T	Me	thoxychlor			Ethyl	
Silvex	0,	P-DDT			iso-Propyl	
MCPA	Р,	P-DDT			n-Propyl	
Tordon					iso-Butyl	
Dicamba	PH	OSPHOROUS PESTI	CIDES		n-Butyl	
	Ro	nnel			n-Amyl	
CHLORINATED PESTICI	DES Pa	rathion		2,4	,5-T Methyl	
Lindane	Me	thyl Parathion			Ethyl	
Heptachlor	Ma	lathion			iso-Propyl	
Aldrin	Di	azinon			n-Propyl	
Heptachlor Epoxide	Et	hion			iso-Butyl	
DDE	Me	thyl Trithion			n-Butyl	
DDD	Ph	orate		<del></del>	n-Amyl	
Dieldrin		PCB	***************************************			
Trace is defined as	a quantity le	ss than 0.1/ug/J	, but s	still dete	ctable. All	concentra
tions are in ug/l of						00110011024
REMARKS: 2 phosph	orous unknown	trace peaks.				
.05 ug/l	Cl of unknown	Chlorinated Pes	ticide	(1 peak)	plus 3 trace	peaks
_		sters (3 peaks)	·			,
	May 4, 1971		Certif	ied:	(IP)	
Form LHP-0012			,		<del>''</del>	

ATTENTION:	E.	E.	Kupchanko	
ATTENTION:	E.	E.	Kupchanko	

Lab. Sample No. 71P -	54		
Type of Sample: Wa	ter	Time	e:
Source of Sample: N	orth Saskatchewan River - 1	.05th Bridge	
Date Sampled: May	21/71	Date Receive	d: <u>May 21/71</u>
ACIDS			ESTERS
2, 4-D	Endrin		2, 4-D Methyl
2,4,5-T	Methoxychlor		Ethyl
Silvex	O, P-DDT		iso-Propyl
MCPA	P, P-DDT		n-Propyl
Tordon			iso-Butyl
Dicamba	PHOSPHOROUS PESTI	CIDES	n-Butyl
	Ronnel		n-Amyl
CHLORINATED PESTICIDES	Parathion		2,4,5-T Methyl
Lindane	Methyl Parathion		Ethyl
Heptachlor	Malathion	****	iso-Propyl
Aldrin	Diazinon		n-Propyl
Heptachlor Epoxide	Ethion		iso-Butyl
DDE	Methyl Trithion		n-Butyl
DDD	Phorate		n-Amyl
Dieldrin	PCB		
Trace is defined as a	quantity less than 0.1/ug/	1, but still	detectable. All concentra-
tions are in mg/l of s	sample.		
REMARKS: Trace peak	of unknown Chlorinated Pes	ticides	
4.82 /ug/1	Cl of unknown Esters -7 pe	aks plus	
3 trace pe	aks ; 5 phosphorous trac	e peaks	
Date:June 29/71		_ Certified:	

ATTENTION:	E. E. Kupchanko	
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Source of Sample:	North S	th Saskatchewan River - Vinca				
_			Date Received: May 21/71			
ACIDS				ESTERS		
2, 4-D		Endrin		_ 2, 4-D Met	hyl	
2-4,5-T		Methoxychlor		Eth	yl	
Silvex		O, P-DDT		_ iso-Pro	pyl	
MCPA .		P, P-DDT	<del></del>	n-Pro	pyl	
Tordon .				iso-But	yl	
Dicamba		PHOSPHOROUS PESTI	CIDES	n-But	yl	
		Ronnel		n-Amy	1	
CHLORINATED PESTICI	DES	Parathion		2,4,5-T Me	thyl	
Lindane		Methyl Parathion		Eth	yl	
Heptachlor .		Malathion		_ iso-Pro	opyl	
Aldrin		Diazinon		_ n-Pro	opyl	
leptachlor Epoxide		Ethion		_ iso-But	tyl	
DDE		Methyl Trithion		n-But	tyl	
ממס		Phorate		n-Amy	/l	
ieldrin		PCB		•		
race is defined as	a quantity	less than 0.1/ug/]	, but still	detectable.	All conce	
ions are in ug/l of						
EMANUO III	, ,	<b>a.</b>				
		wn Chlorinated Pest				
•		nown Esters -6 p				
2 trace	peaks ·	6 phosphorous	trace peaks			